

# Hydrazone-Hydrazones in the Synthesis of 1,3,4-Oxadiazine, 1,2,4-Triazine and Pyrazole Derivatives with Antitumor Activities

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**Abstract:** The reaction of cyanoacetyl hydrazine (**1**) with the  $\alpha$ -haloketones **2a-2c** in 1,4-dioxane afforded the hydrazone-hydrazone derivatives **3a-3c**. The latter products were used in a series of heterocyclization to give 1,3,4-oxadiazine, 1,2,4-triazine and pyrazole derivatives. All compounds synthesized through this work are new products and their antitumor evaluation towards three human tumor cell lines representing different tumor types, namely, breast adenocarcinoma (MCF-7), non-small cell lung cancer (NCI-H460) and CNS cancer (SF-268). Some of the synthesized products showed high inhibitory effect towards the three cell lines.

**Keywords:** Antitumor, hydrazone-hydrazone, 1,3,4-oxadiazine, 1,2,4-triazine, pyrazole.

## INTRODUCTION

It is well known that the hydrazone group plays an important role for the antimicrobial activity. Thus, a number of hydrazone-hydrazone possessed interesting antibacterial-antifungal [1-4], anticonvulsant [5-7], anti-inflammatory [8,9], antimalarial [10] and antituberculosis activities [11-17]. With the aim of obtaining new hydrazone-hydrazones with such wide spectrum of pharmaceutical applications, we report here the synthesis of a series of hydrazone-hydrazones **3a-3c** via the reaction of cyanoacetylhydrazine **1** with  $\alpha$ -bromoketones (**2a-2c**). Moreover, the synthesized hydrazone-hydrazone derivatives were used in a series of heterocyclic transformations to give 1,3,4-oxadiazine, 1,2,4-triazine and pyrazole derivatives [18-21].

## RESULTS AND DISCUSSION

The reaction of cyanoacetyl hydrazine **1** with the  $\alpha$ -haloketones **2a-2c** in 1,4-dioxane at room temperature gave the  $\alpha$ -bromohydrazone derivatives **3a-3c** (Scheme 1). The structures of the latter were elucidated through analytical and spectral data. Thus, the <sup>1</sup>H NMR spectrum of **3a** showed the presence of two singlets at 4.77, 5.38 ppm corresponding to two CH<sub>2</sub> groups, a multiplet at 7.27-7.40 ppm corresponding to phenyl protons and a singlet at 8.88 ppm (D<sub>2</sub>O exchangeable) for an NH group. Moreover, the <sup>13</sup>C NMR of **3a** showed the presence of  $\delta$ : 27.0, 64.6 corresponding to the two CH<sub>2</sub> groups together with other signals which are consistent with the proposed structure (see experimental section).

Compounds **3a-3c** underwent ready cyclization when heated in a boiling water bath with sodium ethoxide in ethanol solution to form the 1,3,4-oxadiazine derivatives **4a-4c** (Scheme 2). The analytical and spectral data obtained for the synthesized compounds were in agreement with the proposed structures (see experimental section).

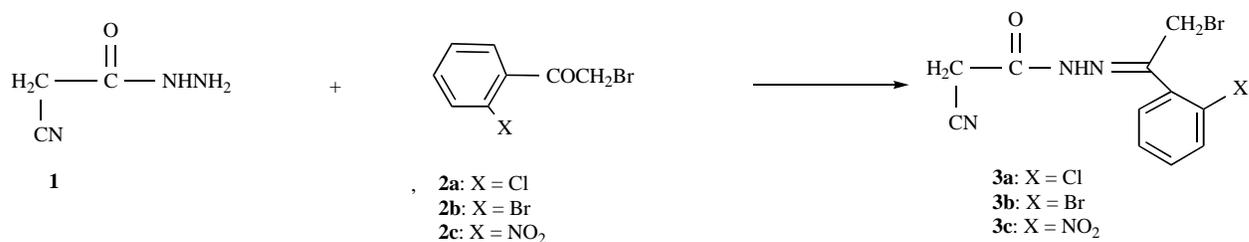
The reaction of **3a** with either hydrazine hydrate **5a** or phenylhydrazine **5b** gave the 1,2,4-triazine derivatives **7a** or **7b**, respectively (Scheme 3). Formation of the latter is explained in terms of the non-isolable intermediate formation of **6a** or **6b** followed by dehydration. The elemental analysis and <sup>1</sup>H NMR spectra were the basis of structure elucidation.

The 1,2,4-triazine derivatives **7a** and **7b** bearing the cyanomethyl group showed a high reactivity towards aromatic aldehydes. Thus, the reaction of compounds **7a** and **7b** with benzaldehyde **8** gave the derivatives **9a** and **9b** (Scheme 4). On the other hand, their reaction with salicylaldehyde **10** gave the coumarin derivatives **11a** and **11b**.

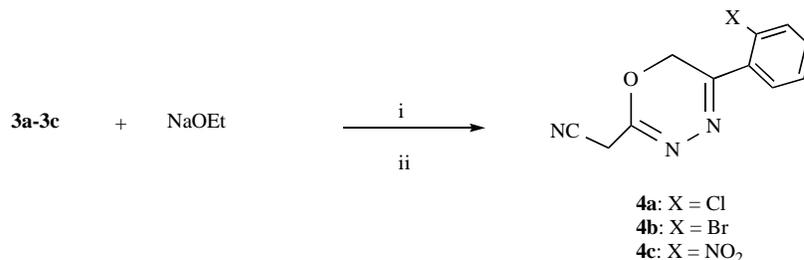
The reaction of compounds **3a-3c** with benzaldehyde gave the 2-benzalcyanomethyl-1,3,4-oxadiazine derivatives **13a-13c** (Scheme 5). The structures of the obtained products were elucidated on the basis of analytical and spectral data. The same compounds **13a-13c**, respectively were obtained through reaction of **4a-4c** with benzaldehyde.

Compounds **3a-3c** bearing the  $\alpha$ -bromomethyl group seemed to be reactive towards nucleophilic reagents. Thus, their reaction with potassium cyanide gave the pyrazole derivatives **15a-15c** (Scheme 6). Formation of these products was based on the formation of the non isolable intermediates **14a-14c** followed by cyclization. Our trials to isolate the acyclic intermediates failed. The analytical and spectral data of compounds **15a-15c** were in agreement with the proposed structures. Thus, the <sup>1</sup>H NMR spectrum of **15a** showed a

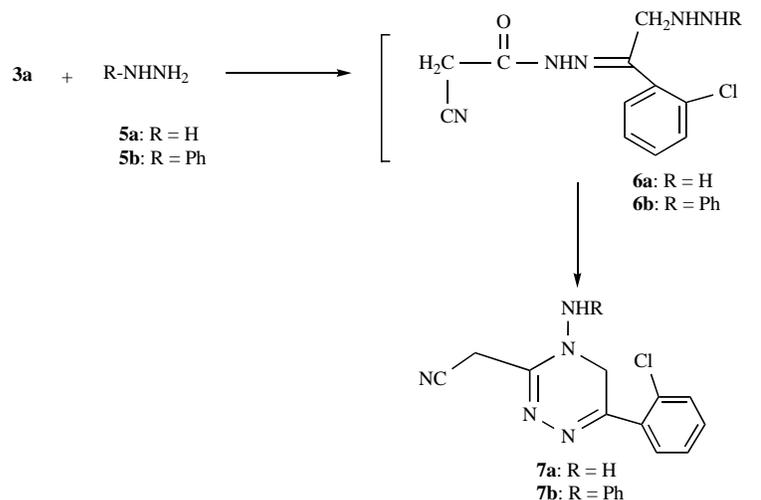
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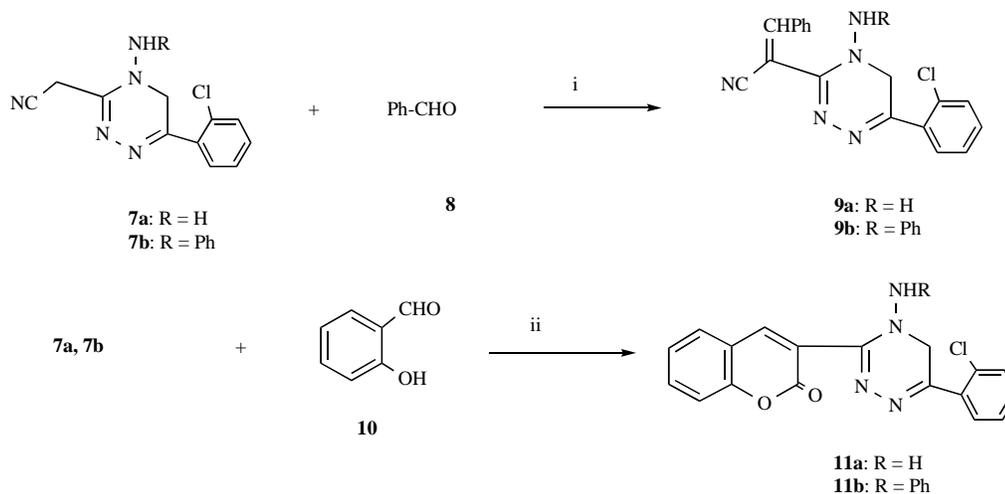
**Scheme 1.** Reagents and conditions: 1,4-dioxane (0.01 mole equiv.), heat 2h, stirring at r.t. 2 h, yield 80% (**3a**), 72 % (**3b**), 88 % (**3c**).



**Scheme 2.** Reagents and conditions: (i) NaOEt, EtOH (0.01 equiv.), heat in a boiling water bath 4 h (ii) HCl in ice/water, pH 7, 70 % (**4a**), 78 % (**4b**), 63 % (**4c**).

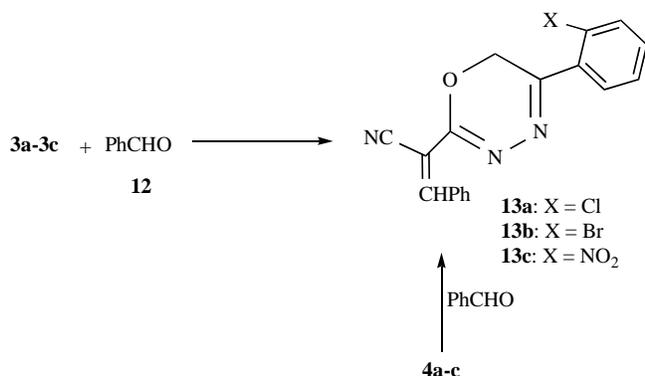


**Scheme 3.** Reagents and conditions: 1,4-dioxane (0.01 equiv.), heat 3 h, ice/water with HCl, 60 % (**7a**), 68 % (**7b**).

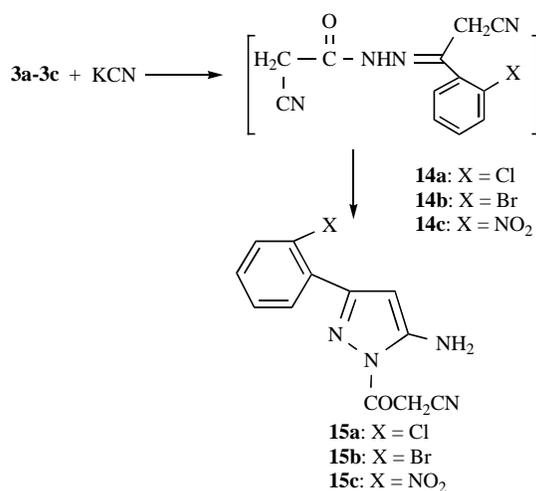


**Scheme 4.** Reagents and conditions: (i) NH<sub>4</sub>OAc, Oil bath 120 °C, 15 min. (0.01 equiv.), EtOH, 72 % (**9a**), 69 % (**9b**). (ii) 1,4-dioxane (0.01 equiv.), piperidine, heat 4 h, EtOH, 62 % (**11a**), 57 % (**11b**).

singlet at 4.64 ppm corresponding to CH<sub>2</sub> group, a singlet at 4.90 ppm corresponding to an NH<sub>2</sub> group, a singlet at 6.52 ppm corresponding to the pyrazolic proton H-4 and a multiplet at 7.28-7.34 ppm for the protons of C<sub>6</sub>H<sub>4</sub> group. Moreover, the <sup>13</sup>C NMR showed signals that are in agreement with the structure of **15a** (see experimental section).



**Scheme 5.** Reagents and conditions: 1,4-dioxane (0.01 mole equiv.), piperidine, heat 6 h, HCl in ice/water, 63 % (**13a**), 66 % (**13b**), 70 % (**13c**).



**Scheme 6.** Reagents and conditions: EtOH (0.01 mole equiv.), water bath at 60 °C, 0.5 h, HCl in ice/water pH 6, 74 % (**15a**), 70 % (**15b**), 64 % (**15c**).

### Effect of the Synthesized Compounds on the Growth of Human Tumor Cell Lines

The effect of compounds **3a-15c** was evaluated on the *in vitro* growth of three human tumor cell lines representing different tumor types, namely, breast adenocarcinoma (MCF-7), non-small cell lung cancer (NCI-H460) and CNS cancer (SF-268), after a continuous exposure of 48 h. The results are summarized in [Table 1](#).

All the tested compounds were able to inhibit the growth of the human tumor cell lines in a dose-dependent manner (data not shown). The 1,3,4-oxadiazine derivative **4c**, the 1,2,4-triazine derivative **11a** and the pyrazole derivative **15c** showed the highest inhibitory effect of the tested compounds, exhibiting an equivalent potency in all the three tumor cell lines. However, their activity is largely lesser than the positive control. While compounds **3b**, **3c**, **7a**, **9a**, **13b**,

**15a** showed moderated growth inhibitory effect relative to compounds **4c**, **11a** and **15c**. On the other hand, compounds **3a**, **4a**, **4b**, **9b**, **11b**, **13a** and **15b** showed the lowest inhibitory effect relative to the tested compounds. It is convenient to observe that the pyrazole derivative **15c** with its *o*-nitrophenyl group showed the maximum inhibitory effect through the three cell lines. Comparing the activities of compounds **4a**, **4b** and **4c**, it is observed that the *o*-chlorophenyl-1,3,4-oxazine **4a** presents a weaker growth inhibitory effect while the *o*-nitrophenyl-1,3,4-oxazine **4c** showed high inhibitory effect although the results in NCI-H460 cell line are comparable.

## EXPERIMENTAL SECTION

### General Information

Melting points were determined in open capillaries and are uncorrected. I.R spectra were measured using KBr discs on a Pye Unicam SP-1000 spectrophotometer. <sup>1</sup>H NMR & <sup>13</sup>C NMR spectra were measured on a Varian EM390-200 MHz instrument in DMSO-d<sub>6</sub> as solvent and using TMS as internal standard, and chemical shifts (δ) are expressed in ppm. Elemental analyses were determined on a Yanaco CHN Corder elemental analyzer (Japan).

### *α*-Bromo(*o*-chloroacetophenone)-*α*-cyanoacetylhydrazone **3a**, *β*-bromo(*o*-bromoacetophenone)-*α*-cyanoacetylhydrazone **3b** and *β*-bromo(*o*-nitroacetophenone)-*α*-cyanoacetylhydrazone **3c**

To a solution of cyanoacetyl hydrazine **1** (0.99 g, 0.01 mol), in 1,4-dioxane (40 mL) either of *o*-chlorophenacylbromide **2a** (2.3 g, 0.01 mol), *o*-bromophenacylbromide **2b** (2.80 g, 0.01 mol) or *o*-nitrophenacylbromide (2.46 g, 0.01 mol) was added. The reaction mixture was kept at room temperature with stirring for 2 h and the formed solid product was filtrated off.

Compound **3a**: Pale yellow crystals were recovered from ethanol in a yield of 80 % (2.49 g); mp: 177-179°C. I.R (ν/cm<sup>-1</sup>) = 3480-3334 (NH), 3054 (CH aromatic), 2887 (CH<sub>2</sub>), 2260 (CN), 1684 (C=O), 1666 (C=N), 1641 (C=C). <sup>1</sup>H NMR (200 MHz, DMSO-d<sub>6</sub>) δ 4.77 (s, 2H, CH<sub>2</sub>), 5.38 (s, 2H, CH<sub>2</sub>), 7.27-7.40 (m, 4H, C<sub>6</sub>H<sub>4</sub>), 8.88 (s, 1H, NH). <sup>13</sup>C NMR (200 MHz, DMSO-d<sub>6</sub>) δ 27.0 (CH<sub>2</sub>), 64.6 (CH<sub>2</sub>), 116.3 (CN), 126.8, 127.0, 128.7, 130.0, 138.4 (C<sub>6</sub>H<sub>5</sub>), 157.3 (C=N), 172.7 (C=O). Calculated for C<sub>11</sub>H<sub>9</sub>BrClN<sub>3</sub>O: 314.57, C: 42.00, H: 2.88, N: 13.36). Found, C: 42.09, H: 3.11, N: 13.40.

Compound **3b**: Pale yellow crystals were recovered from ethanol in a yield of 72 % (2.59 g); mp: 202-205°C. I.R (ν/cm<sup>-1</sup>) = 3463-3329 (NH), 3050 (CH aromatic), 2887 (CH<sub>2</sub>), 2260 (CN), 1690 (C=O), 1666 (C=N), 1636 (C=C). <sup>1</sup>H NMR (200 MHz, DMSO-d<sub>6</sub>) δ 4.74 (s, 2H, CH<sub>2</sub>), 5.40 (s, 2H, CH<sub>2</sub>), 7.26-7.35 (m, 4H, C<sub>6</sub>H<sub>4</sub>), 8.83 (s, 1H, NH). <sup>13</sup>C NMR (200 MHz, DMSO-d<sub>6</sub>) δ 27.2 (CH<sub>2</sub>), 64.5 (CH<sub>2</sub>), 116.0 (CN), 126.0, 126.8, 127.0, 128.7, 133.0, 139.4 (C<sub>6</sub>H<sub>5</sub>), 157.0 (C=N), 172.9 (C=O). Calculated for C<sub>11</sub>H<sub>9</sub>Br<sub>2</sub>N<sub>3</sub>O: 359.02, C: 36.80, H: 2.53, N: 11.70. Found, C: 36.66, H: 2.49, N: 11.38.

Compound **3c**: Orange crystals were recovered from ethanol in a yield of 88 % (2.84 g); mp: 210-214°C. I.R (ν/cm<sup>-1</sup>) = 3465-3326 (NH), 3056 (CH aromatic), 2888

Table 1. Effect of Compounds 3-15c on the Growth of Three Human Tumor Cell Lines

Compound No.	GI <sub>50</sub> (μM)		
	MCF-7	NCI-H460	SF-268
3a	66.6 ± 12.2	12 ± 6.2	24.8 ± 3.2
3b	20 ± 0.4	24.3 ± 0.8	32 ± 0.8
3c	30 ± 0.6	17.3 ± 1.4	22.3 ± 1.5
4a	40.6 ± 12.6	32.6 ± 8.6	60.4 ± 14.8
4b	72.7 ± 17.5	40.2 ± 12.8	50.0 ± 9.01
4c	11.8 ± 0.6	14.5 ± 0.8	16.7 ± 1.6
7a	35.4 ± 10.2	24.1 ± 0.8	18.9 ± 6.8
7b	38.0 ± 1.8	44.0 ± 0.8	20.5 ± 1.1
9a	22.0 ± 0.2	30.6 ± 1.4	38.4 ± 0.6
9b	50.1 ± 0.7	23.2 ± 4.8	18.4 ± 1.8
11a	11.9 ± 0.5	14.1 ± 0.6	20.3 ± 0.5
11b	70.9 ± 0.9	43.6 ± 1.8	56.8 ± 0.8
13a	66.6 ± 16.9	38.9 ± 10.8	50.8 ± 8.6
13b	40.6 ± 12.2	32.6 ± 8.6	60.4 ± 14.8
13c	22.0 ± 0.2	30.6 ± 1.4	38.4 ± 0.6
15a	20 ± 0.4	24.3 ± 0.8	32 ± 0.8
15b	70.9 ± 0.9	43.6 ± 1.8	56.8 ± 0.8
15c	2.5 ± 0.5	10.4 ± 0.6	8.0 ± 0.4
Doxorubicin	0.0428 ± 0.008	0.0940 ± 0.008	0.0940 ± 0.007

Results are given in concentrations that were able to cause 50 % of cell growth inhibition (GI<sub>50</sub>) after a continuous exposure of 48 h and show means ± SEM of three-independent experiments performed in duplicate.

(CH<sub>2</sub>), 2263 (CN), 1685 (C=O), 1661 (C=N), 1634 (C=C). <sup>1</sup>H NMR (200 MHz, DMSO-d<sub>6</sub>) δ 4.79 (s, 3H, CH<sub>2</sub>), 5.44 (s, 2H, CH<sub>2</sub>), 7.31-7.39 (m, 4H, C<sub>6</sub>H<sub>4</sub>), 8.80 (s, 1H, NH). <sup>13</sup>C NMR (200 MHz, DMSO-d<sub>6</sub>) δ 27.1 (CH<sub>2</sub>), 64.6 (CH<sub>2</sub>), 116.6 (CN), 126.2, 127.6, 128.9, 130.8, 138.1 (C<sub>6</sub>H<sub>5</sub>), 157.2 (C=N), 172.5 (C=O). . Calculated for C<sub>11</sub>H<sub>9</sub>BrN<sub>4</sub>O<sub>3</sub>: 323.99, C: 40.64, H: 2.79, N: 17.23. Found, C: 40.40, H: 3.09, N: 17.11.

**5-(p-Clorophenyl)-2-cyanomethyl-1,3,4-oxadiazine 4a** **5-(p-bromo-phenyl)-2-acetonitrilo-1,3,4-oxadiazine 4b** and **5-(p-nitrophenyl)-2-cyanomethyl-1,3,4-oxadiazine 4c**

A suspension of either **3a** (3.12 g, 0.01 mol), **3b** (3.65 g, 0.01 mol) or **3c** (3.24 g, 0.01 mol) in sodium ethoxide [prepared by dissolving sodium metal (0.46 g, 0.02 mol) in absolute ethanol (25 mL)] was heated in a boiling water bath for 4 h then left to cool. The solid product formed upon pouring into ice/water containing hydrochloric acid (till pH 7) was collected by filtration.

**Compound 4a:** Pale yellow crystals were recovered from 1,4-dioxane in a yield of 70 % (1.63 g); mp: 262-265°C. I.R (ν/cm<sup>-1</sup>) = 3060 (CH aromatic), 2870 (CH<sub>2</sub>), 2245 (CN), 1660 (C=N), 1638 (C=C). <sup>1</sup>H NMR (200 MHz, DMSO-d<sub>6</sub>) δ 4.65 (s, 2H, CH<sub>2</sub>), 5.89 (s, 2H, ring CH<sub>2</sub>), 7.30-7.38 (m,

4H, C<sub>6</sub>H<sub>4</sub>). <sup>13</sup>C NMR (200 MHz, DMSO-d<sub>6</sub>) δ 19.8 (CH<sub>2</sub>), 64.6 (ring CH<sub>2</sub>), 116.8 (CN), 126.0, 127.3, 129.9, 130.8, 138.0 (C<sub>6</sub>H<sub>5</sub>), 164.2, 164.8 (2 C=N). Calculated for C<sub>11</sub>H<sub>8</sub>ClN<sub>3</sub>O: 233.04, C: 56.54, H: 3.45, N: 17.98. Found, C: 56.51, H: 3.60, N: 18.02.

**Compound 4b:** Pale orange crystals were recovered from 1,4-dioxane in a yield of 78 % (2.17 g); mp: 266-269°C. I.R (ν/cm<sup>-1</sup>) = 3066 (CH aromatic), 2880 (CH<sub>2</sub>), 2222 (CN), 1660 (C=N), 1635 (C=C). <sup>1</sup>H NMR (200 MHz, DMSO-d<sub>6</sub>) δ 4.77 (s, 2H, CH<sub>2</sub>), 5.73 (s, 2H, ring CH<sub>2</sub>), 7.31-7.40 (m, 4H, C<sub>6</sub>H<sub>4</sub>). <sup>13</sup>C NMR (200 MHz, DMSO-d<sub>6</sub>) δ 19.8 (CH<sub>2</sub>), 64.6 (ring CH<sub>2</sub>), 116.6 (CN), 126.3, 127.0, 129.6, 130.4, 138.2 (C<sub>6</sub>H<sub>5</sub>), 164.0, 164.6 (2 C=N). Calculated for C<sub>11</sub>H<sub>8</sub>BrN<sub>3</sub>O: 278.12, C: 47.51, H: 2.90, N: 15.11. Found, C: 47.44, H: 3.30, N: 14.92.

**Compound 4c:** Pale orange crystals were recovered from 1,4-dioxane in a yield of 63 % (1.53g); mp: 246-249°C. I.R (ν/cm<sup>-1</sup>) = 3066 (CH aromatic), 2871 (CH<sub>2</sub>), 2226 (CN), 1660 (C=N), 1639 (C=C). <sup>1</sup>H NMR (200 MHz, DMSO-d<sub>6</sub>) δ 4.77 (s, 2H, CH<sub>2</sub>), 5.80 (s, 2H, ring CH<sub>2</sub>), 7.30-7.38 (m, 4H, C<sub>6</sub>H<sub>4</sub>). Calculated for C<sub>11</sub>H<sub>8</sub>N<sub>4</sub>O<sub>3</sub>: 244.06, C: 54.10, H: 3.30, N: 22.94. Found, C: 53.89, H: 3.36, N: 23.42.

**3-cyanomethyl-4-amino-6-phenyl-1,2,4-triazine 7a and 3-cyanomethyl-4-phenylamino-6-phenyl-1,2,4-triazine 7b**

General procedure: To a solution of **3a** (3.12 g, 0.01 mol) in 1,4-dioxane (40 mL) either hydrazine hydrate (0.50 g, 0.01 mol) or phenylhydrazine (1.08 g, 0.01 mol) was added. The reaction mixture was heated under reflux for 3 h then left to cool. The solid product formed upon pouring onto ice/water containing few drops of hydrochloric acid was collected by filtration.

Compound **7a**: Yellow crystals were recovered from 1,4-dioxane in a yield of 60 % (1.48 g); mp: 164-166°C. IR ( $\nu/\text{cm}^{-1}$ ) = 3059 (CH aromatic), 2890 (CH<sub>2</sub>), 2225 (CN), 1660 (C=N), 1635 (C=C). <sup>1</sup>H NMR (200 MHz, DMSO-d<sub>6</sub>)  $\delta$ 4.46 (s, 2H, CH<sub>2</sub>), 4.87 (s, 2H, NH<sub>2</sub>), 5.66 (s, 2H, triazine CH<sub>2</sub>), 7.28-7.34 (m, 4H, C<sub>6</sub>H<sub>4</sub>). <sup>13</sup>C NMR (200 MHz, DMSO-d<sub>6</sub>)  $\delta$ 19.7 (CH<sub>2</sub>), 64.6 (ring CH<sub>2</sub>), 116.4 (CN), 126.8, 127.3, 129.5, 129.9, 138.0 (C<sub>6</sub>H<sub>5</sub>), 164.3, 164.8 (2 C=N). Calculated for C<sub>11</sub>H<sub>10</sub>ClN<sub>5</sub>: 247.06, C: 53.34, H: 4.07, N: 28.28. Found, C: 53.79, H: 4.09, N: 28.39.

Compound **7b**: Yellow crystals were recovered from 1,4-dioxane in a yield of 68 % (2.19 g); mp: 198 °C. IR ( $\nu/\text{cm}^{-1}$ ) = 3055 (CH aromatic), 2883 (CH<sub>2</sub>), 2221 (CN), 1662 (C=N), 1637 (C=C). <sup>1</sup>H NMR (200 MHz, DMSO-d<sub>6</sub>)  $\delta$ 4.47 (s, 2H, CH<sub>2</sub>), 4.83 (s, 2H, NH<sub>2</sub>), 5.68 (s, 2H, triazine CH<sub>2</sub>), 7.33-7.38 (m, 4H, C<sub>6</sub>H<sub>4</sub>), 8.37 (s, 1H, NH). <sup>13</sup>C NMR (200 MHz, DMSO-d<sub>6</sub>)  $\delta$ 19.7 (CH<sub>2</sub>), 64.6 (ring CH<sub>2</sub>), 116.4 (CN), 126.8, 127.3, 129.5, 129.9, 138.0 (C<sub>6</sub>H<sub>5</sub>), 164.3, 164.8 (2 C=N). Calculated for C<sub>17</sub>H<sub>14</sub>ClN<sub>5</sub>: 323.09, C: 63.06, H: 4.36, N: 21.63. Found, C: 62.88, H: 4.86, N: 21.72.

**3- $\alpha$ -Benzalacetoneitrilo-4-amino-6-(p-chlorophenyl)-1,2,4-triazine 9a, and 3- $\alpha$ -benzalacetoneitrilo-4-phenylamino-6-(p-chlorophenyl)-1,2,4-triazine 9b**

General Procedure: To an equimolecular amounts of dry solid of either **7a** (2.47 g, 0.01 mol) or **7b** (3.23 g, 0.01 mol), benzaldehyde (1.08 g, 0.01 mol) and ammonium acetate (1.0 g) were added. The whole reaction mixture was heated in an oil bath at 120 °C for 15 min then left to cool. The solid product formed upon triturating the remaining product with ethanol was filtered off.

Compound **9a**: Pale yellow crystals were recovered from acetic acid in a yield of 72 % (2.41 g); mp: 190-193°C. IR ( $\nu/\text{cm}^{-1}$ ) = 3056 (CH aromatic), 2873 (CH<sub>2</sub>), 2220 (CN), 1653 (C=N), 1639 (C=C). <sup>1</sup>H NMR (200 MHz, DMSO-d<sub>6</sub>)  $\delta$ 4.99 (s, 2H, NH<sub>2</sub>), 5.64 (s, 2H, triazine CH<sub>2</sub>), 6.61 (s, 1H, CH=C), 147.2 (CH=C), 7.30-7.39 (m, 9H, C<sub>6</sub>H<sub>5</sub>, C<sub>6</sub>H<sub>4</sub>). <sup>13</sup>C NMR (200 MHz, DMSO-d<sub>6</sub>)  $\delta$ 19.7 (CH<sub>2</sub>), 64.6 (ring CH<sub>2</sub>), 116.8 (CN), 96.9 (CH=C), 145.7 (CH=C), 126.6, 127.0, 129.1, 129.9, 130.5, 132.8, 138.8 (C<sub>6</sub>H<sub>5</sub>, C<sub>6</sub>H<sub>4</sub>), 164.0, 164.9 (2 C=N). Calculated for C<sub>18</sub>H<sub>14</sub>ClN<sub>5</sub>: 335.09, C: 64.38, H: 4.20, N: 20.86. Found, C: 64.54, H: 4.49, N: 21.08.

Compound **9b**: Orange crystals were recovered from 1,4-dioxane in a yield of 69 % (2.83 g); mp: 180-183°C. IR ( $\nu/\text{cm}^{-1}$ ) = 3058 (CH aromatic), 2876 (CH<sub>2</sub>), 2219 (CN), 1660 (C=N), 1643 (C=C). <sup>1</sup>H NMR (200 MHz, DMSO-d<sub>6</sub>)  $\delta$ 5.74 (s, 2H, triazine CH<sub>2</sub>), 6.33 (s, 1H, CH=C), 7.27-7.38 (m, 14H, 2C<sub>6</sub>H<sub>5</sub>, C<sub>6</sub>H<sub>4</sub>), 8.21 (s, 1H, NH). <sup>13</sup>C NMR (200 MHz, DMSO-d<sub>6</sub>)  $\delta$ 19.9 (CH<sub>2</sub>), 64.5 (ring CH<sub>2</sub>), 116.2 (CN), 96.3 (CH=C), 145.6 (CH=C), 126.6, 127.3, 129.0, 129.9, 130.2, 131.7, 134.0, 136.9, 138.6 (2C<sub>6</sub>H<sub>5</sub>, C<sub>6</sub>H<sub>4</sub>), 164.2,

164.3 (2 C=N). Calculated for C<sub>24</sub>H<sub>18</sub>ClN<sub>5</sub>: 411.13, C: 69.98, H: 4.40, N: 17.00. Found, C: 70.24, H: 4.66, N: 16.78.

**3-(Coumarin-3-yl)-4-amino-6-(p-chlorophenyl)-1,2,4-triazine 11a and 3-(coumarin-3-yl)-4-phenylamino-6-(p-chlorophenyl)-1,2,4-triazine 11a**

To a solution of either **7a** (2.47 g, 0.01 mol) or **7b** (3.23 g, 0.01 mol) in 1,4-dioxane (40 mL) containing piperidine (0.5 mL), salicylaldehyde (1.22 g, 0.01 mol) was added. The reaction mixture was heated under reflux for 4 h then evaporated under vacuum. The remaining product was triturated with ethanol and the solidified product was collected by filtration.

Compound **11a**: Pale yellow crystals were recovered from acetic acid in a yield of 62 % (2.18 g); mp: 180-184°C. IR ( $\nu/\text{cm}^{-1}$ ) = 3060 (CH aromatic), 1690 (CO), 1673 (C=N), 1644 (C=C). <sup>1</sup>H NMR (200 MHz, DMSO-d<sub>6</sub>)  $\delta$ 4.67 (s, 2H, NH<sub>2</sub>), 5.81 (s, 2H, triazine CH<sub>2</sub>), 6.88 (s, 1H, coumarin H-4), 7.32-7.39 (m, 8H, 2C<sub>6</sub>H<sub>4</sub>). <sup>13</sup>C NMR (200 MHz, DMSO-d<sub>6</sub>)  $\delta$ 60.5 (ring CH<sub>2</sub>), 96.0 (coumarin H-4), 120.3, 122.6, 124.0, 125.9, 127.0, 129.3, 129.9, 130.2, 131.7, 134.0, 136.9, 138.0 (2 C<sub>6</sub>H<sub>4</sub>), 159.0 (C=O), 164.3, 164.6 (2 C=N). Calculated for C<sub>18</sub>H<sub>13</sub>ClN<sub>4</sub>O<sub>2</sub>: 354.08, C: 61.28, H: 3.71, N: 15.88. Found, C: 61.52, H: 4.06, N: 16.45.

Compound **11b**: Pale brown crystals were recovered from acetic acid in a yield of 57 % (2.43 g); mp: 210-213°C. IR ( $\nu/\text{cm}^{-1}$ ) = 3055 (CH aromatic), 1688 (CO), 1665 (C=N), 1644 (C=C). <sup>1</sup>H NMR (200 MHz, DMSO-d<sub>6</sub>)  $\delta$ 4.42 (s, 2H, NH<sub>2</sub>), 5.77 (s, 2H, triazine CH<sub>2</sub>), 6.90 (s, 1H, coumarin H-4), 7.28-7.38 (m, 8H, 2C<sub>6</sub>H<sub>4</sub>). <sup>13</sup>C NMR (200 MHz, DMSO-d<sub>6</sub>)  $\delta$ 60.0 (ring CH<sub>2</sub>), 96.3 (coumarin H-4), 120.6, 121.9, 123.6, 125.5, 127.4, 128.7, 129.3, 130.0, 131.9, 134.8, 135.7, 138.3, 144.2, 146.9 (C<sub>6</sub>H<sub>5</sub>, 2 C<sub>6</sub>H<sub>4</sub>), 158.8 (C=O), 164.1, 164.5 (2 C=N). Calculated for C<sub>24</sub>H<sub>17</sub>ClN<sub>4</sub>O<sub>2</sub>: 428.10, C: 67.21, H: 4.00, N: 13.06. Found, C: 67.09, H: 3.85, N: 13.39.

**2- $\alpha$ -5-p-chlorophenyl-1,3,4-oxadiazine 13a, 2- $\alpha$ -benzalcyanomethyl-5-p-bromophenyl-1,3,4-oxadiazine 13b and 2- $\alpha$ -benzalcyanomethyl-5-p-nitrophenyl-1,3,4-oxadiazine (13c)**

General procedure: To a solution of either **3a** (3.12 g, 0.01 mol), **3b** (3.65 g, 0.01 mol), **3c** (3.24 g, 0.01 mol), **4a** (2.33 g, 0.01 mol), **4b** (2.79 g, 0.01 mol) or **4c** (2.44 g, 0.01 mol) in 1,4-dioxane (40 mL) containing piperidine (1.0 mL) benzaldehyde (1.08 g, 0.01 mol) was added. The reaction mixture was heated under reflux for 6 h then poured onto ice/water containing few drops of hydrochloric acid. The solid product, formed in each case, was collected by filtration to afford the respective products.

Compound **13a**: Yellow crystals were recovered from methanol in a yield of 63 % (2.02 g), from **3a** and 70 % (2.25) from **4a**; mp: 180-182°C. IR ( $\nu/\text{cm}^{-1}$ ) = 3055 (CH aromatic), 2220 (CN), 1656 (C=N), 1640 (C=C). <sup>1</sup>H NMR (200 MHz, DMSO-d<sub>6</sub>)  $\delta$ 5.60 (s, 2H, oxadiazine CH<sub>2</sub>), 6.30 (s, 1H, CH=C), 7.30-7.39(m, 9H, C<sub>6</sub>H<sub>5</sub>, C<sub>6</sub>H<sub>4</sub>). <sup>13</sup>C NMR (200 MHz, DMSO-d<sub>6</sub>)  $\delta$ 60.4 (ring CH<sub>2</sub>), 95.0 (CH=C), 118.0 (CH=C), 120.3, 121.5, 123.2, 125.0, 127.4, 129.3, 130.0, 134.8, 135.0, 144.0 (C<sub>6</sub>H<sub>5</sub>, C<sub>6</sub>H<sub>4</sub>), 164.0, 164.2 (2 C=N). Calculated for C<sub>18</sub>H<sub>12</sub>ClN<sub>3</sub>O: 322.06, C: 67.19, H: 3.76, N: 13.06. Found, C: 67.33, H: 4.06, N: 12.74.

Compound **13b**: Orange-red crystals were recovered from methanol/dioxane mixture in a yield of 66% (2.41 g); mp: 210-212°C. I.R ( $\nu/\text{cm}^{-1}$ ) = 3060 (CH aromatic), 2226 (CN), 1665 (C=N), 1621 (C=C).  $^1\text{H}$  NMR (200 MHz, DMSO- $d_6$ )  $\delta$ 5.77 (s, 2H, oxadiazine  $\text{CH}_2$ ), 6.20 (s, 1H, CH=C), 7.29-7.39 (m, 9H,  $\text{C}_6\text{H}_5$ ,  $\text{C}_6\text{H}_4$ ).  $^{13}\text{C}$  NMR (200 MHz, DMSO- $d_6$ )  $\delta$ 60.6 (ring  $\text{CH}_2$ ), 95.2 (CH=C), 118.3 (CH=C), 120.0, 120.2, 121.6, 123.2, 127.4, 128.0, 130.2, 132.6, 134.2, 140.4 ( $\text{C}_6\text{H}_5$ ,  $\text{C}_6\text{H}_4$ ), 164.1, 164.8 (2 C=N). Calculated for  $\text{C}_{18}\text{H}_{12}\text{BrN}_3\text{O}$ : 365.02, C: 59.03, H: 3.30, N: 11.47. Found, C: 58.92, H: 3.55, N: 11.51.

Compound **13c**: Orange crystals were recovered from 1,4-dioxane in a yield of 70 % (2.32 g); mp: 233-236°C. I.R ( $\nu/\text{cm}^{-1}$ ) = 3063 (CH aromatic), 2223 (CN), 1656 (C=N), 1638 (C=C).  $^1\text{H}$  NMR (200 MHz, DMSO- $d_6$ )  $\delta$ 5.80 (s, 2H, ring  $\text{CH}_2$ ), 6.21 (s, 1H, CH=C), 7.32-7.46 (m, 9H,  $\text{C}_6\text{H}_4$ ).  $^{13}\text{C}$  NMR (200 MHz, DMSO- $d_6$ )  $\delta$ 60.3 (ring  $\text{CH}_2$ ), 95.5 (CH=C), 118.1 (CH=C), 120.0, 120.5, 121.6, 123.2, 125.9, 126.8, 130.8, 132.9, 133.9, 141.0 ( $\text{C}_6\text{H}_5$ ,  $\text{C}_6\text{H}_4$ ), 164.3, 164.5 (2 C=N). MS:  $m/e = 332$  ( $\text{M}^+$ ). Calculated for  $\text{C}_{18}\text{H}_{12}\text{N}_4\text{O}_3$ : 332.09, C: 65.06, H: 3.64, N: 16.86. Found, C: 64.88, H: 4.42, N: 16.55.

**5-Amino-1-cyanoacetyl-3-(p-chlorophenyl)-pyrazole 15a**,  
**5-Amino-1-cyanoacetyl-3-(p-bromophenyl)-pyrazole 15b**  
**and 5-Amino-1-cyanoacetyl-3-(p-nitrophenyl)-pyrazole 15c**

General procedure: To a solution of either **3a** (3.12 g, 0.01 mol), **3b** (3.65 g, 0.01 mol) or **3c** (3.24 g, 0.01 mol) in ethanol (40 mL), a solution of potassium cyanide (2.8 g, 0.05 mol) was added. The reaction mixture, in each case, was heated in a warm water bath at 60 °C for 0.5 h then left with stirring at room temperature overnight. The solid product formed upon pouring onto ice/water containing hydrochloric acid (till pH 6) was collected by filtration.

Compound **15a**: Orange crystals were recovered from 1,4-dioxane in a yield of 74 % (1.92 g); mp: 175-178°C. I.R ( $\nu/\text{cm}^{-1}$ ) = 3460-3377 ( $\text{NH}_2$ ), 3050 (CH aromatic), 2250 (CN), 1688 (C=O), 1650 (C=N), 1636 (C=C).  $^1\text{H}$  NMR (200 MHz, DMSO- $d_6$ )  $\delta$ 4.64 (s, 2H,  $\text{CH}_2$ ), 4.90 (s, 2H,  $\text{NH}_2$ ), 6.52 (s, 1H, pyrazole H-4), 7.28-7.34 (m, 4H,  $\text{C}_6\text{H}_4$ ).  $^{13}\text{C}$  NMR (200 MHz, DMSO- $d_6$ )  $\delta$ 60.6 (ring  $\text{CH}_2$ ), 95.2 (CH=C), 118.0 (CH=C), 120.3, 120.9, 121.4, 123.0, 124.9, 126.9, 138.8, 130.9, 133.9, 140.2 ( $\text{C}_6\text{H}_5$ ,  $\text{C}_6\text{H}_4$ ), 164.2, 164.8 (2 C=N). MS:  $m/e = 260$  ( $\text{M}^+$ ). Calculated for  $\text{C}_{12}\text{H}_9\text{ClN}_4\text{O}$ : 260.05, C: 55.29, H: 3.48, N: 21.49. Found, C: 55.43, H: 3.68, N: 21.26.

Compound **15b**: Brown crystals were recovered from 1,4-dioxane in a yield of 70 % (2.13 g); mp: of 177-180°C. I.R ( $\nu/\text{cm}^{-1}$ ) = 3455-3349 ( $\text{NH}_2$ ), 3055 (CH aromatic), 2243 (CN), 1690 (C=O), 1655 (C=N), 1638 (C=C).  $^1\text{H}$  NMR (200 MHz, DMSO- $d_6$ )  $\delta$ 4.65 (s, 2H,  $\text{CH}_2$ ), 4.75 (s, 2H,  $\text{NH}_2$ ), 6.49 (s, 1H, pyrazole H-4), 7.29-7.38 (m, 4H,  $\text{C}_6\text{H}_4$ ).  $^{13}\text{C}$  NMR (200 MHz, DMSO- $d_6$ )  $\delta$ 60.3 (ring  $\text{CH}_2$ ), 95.2 (CH=C), 118.4 (CH=C), 120.6, 120.3, 121.0, 123.3, 124.7, 126.9, 138.8, 130.9, 133.9, 140.9 ( $\text{C}_6\text{H}_5$ ,  $\text{C}_6\text{H}_4$ ), 164.0, 164.8 (2 C=N). Calculated for  $\text{C}_{12}\text{H}_9\text{BrN}_4\text{O}$ : 305.13, C: 47.24, H: 2.97, N: 18.36. Found, C: 47.08, H: 3.22, N: 18.73.

Compound **15c**: Dark orange crystals were recovered from 1,4-dioxane in a yield of 64 % (1.73 g); mp: 195-199 °C. I.R ( $\nu/\text{cm}^{-1}$ ) = 3459-3354 ( $\text{NH}_2$ ), 3052 (CH aromatic), 2239 (CN), 1688 (C=O), 1657 (C=N), 1636 (C=C).  $^1\text{H}$  NMR

(200 MHz, DMSO- $d_6$ )  $\delta$ 4.67 (s, 2H,  $\text{CH}_2$ ), 4.79 (s, 2H,  $\text{NH}_2$ ), 6.52 (s, 1H, pyrazole H-4), 7.29-7.36 (m, 4H,  $\text{C}_6\text{H}_4$ ).  $^{13}\text{C}$  NMR (200 MHz, DMSO- $d_6$ )  $\delta$ 60.3 (ring  $\text{CH}_2$ ), 95.2 (CH=C), 118.0 (CH=C), 120.3, 120.9, 121.4, 123.0, 124.9, 126.9, 138.8, 130.9, 133.9, 138.2 ( $\text{C}_6\text{H}_5$ ,  $\text{C}_6\text{H}_4$ ), 164.5, 164.7 (2 C=N). MS:  $m/e = 271$  ( $\text{M}^+$ ). Calculated for  $\text{C}_{12}\text{H}_9\text{N}_5\text{O}_3$ : 271.07, C: 53.14, H: 3.34, N: 25.82. Found, C: 52.88, H: 3.20, N: 25.43.

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