

# Progress in Plant Polyploidization Based on Antimicrotubular Drugs

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**Abstract:** The results of artificial polyploidization for horticultural and agronomic plant species are presented in this review. The data on use of the classical antimitotic drug colchicine for chromosome doubling are reported. The efficacy of other compounds such as dinitroanilines and phosphorothioamides in polyploidy induction as compared to colchicine has been summarized. The molecular basis of highly specific binding of dinitroanilines and phosphorothioamides with plant tubulin for induction of efficient polyploidy are discussed.

## INTRODUCTION

The rapid advance to genetic fixation which accompanies the doubling of a chromosome set is an attractive feature of haploid breeding, especially in those species where conventional inbreeding involves several years of manual self-pollination. Moreover, it was established that the frequency of spontaneous chromosome doubling in many plant haploids is low and extremely unreliable. To achieve a reproducible and prompt success in plant chromosome doubling there was a need to find efficient, reliable low-cost and safe agents that would promote such effects in plant cells. The experiments on polyploidization of plants were begun in 1940, when colchicine, a natural alkaloid with an antimitotic activity, obtained from the wild plant species *Colchicum autumnale*, was found as an agent with chromosome doubling activity in cells of higher plants [1]. The techniques of colchicine application have not changed much since Levan [2] soaked *Allium* (onion) roots in colchicine solutions.

## COLCHICINE FOR CHROMOSOME DOUBLING

Colchicine has proven to be most useful to double the chromosome numbers of numerous crop species, including decorative flowers, medical and agricultural plants. This alkaloid, however, is very toxic to human beings and also shows undesirable mutagenic activity on plants. A comprehensive review summarizing, nevertheless, the great impact of colchicine in plant polyploidization and its roles in different plant breeding programs was written by Hancock [1]. This author summarized the data where it was shown that this compound was used for induction of polyploidy in fruit and berries, such as apples, cranberries, grapes, peaches, pears and strawberries; for chromosome doubling of several flowers, including marigolds, pinks, snapdragons, petunias,

delphinium and lilies. It was also used for chromosome doubling of some agronomic crops as lettuce, chilli peppers, cotton, potato, rye, sugar beet, wheat and tobacco [1]. By 1979, the chromosome numbers of well over 150 plant species have been doubled using this compound [3]. Unfortunately, only a small number of the artificially induced crops were ever released and few of these autopolyploids ever dominated world market [1].

Colchicine is commonly used in diploidization and polyploidization of different plant species. Currently, apical meristems, secondary buds, tillers or roots are treated with colchicine [see, 4-6]. Generally, about 50% of the treated plants are responsive, but what is important, that three months can be added to the plant regeneration time to recover homozygous lines [7]. Additional drawbacks to using this approach include the regeneration of chimeras in many species [4-5], [8-12], the occurrence of aneuploids [13], abnormalities in plant development [14-15] and low seed yield. Chromosome losses or rearrangements and gene mutations caused by colchicine were found in flax, sunflower, barley and cotton [16]. Such chimaeric plants are unsuitable for breeding since the ploidy of such plants is often unstable through vegetative multiplication.

A more promising alternative is application of colchicine to cultures prior to organ formation (anthers and microspores). The use of haploid plants, generated from anther or microscope culture, has enhanced the efficiency of crop improvement programs [17]. Although haploid plants of different species can be readily regenerated *in vitro*, the haploids cannot be used directly in genetic studies and breeding programs because they are sterile [18].

In particular, colchicine techniques for microspore cultures have established more efficient ways for the production of non-chimaeric doubled haploids [19]. Colchicoids were produced successfully from corn anther-derived callus [12], [20, 21] and microspore culture [22] as well as from wheat anther culture [8, 23-25]. Colchicine diploidization at the microspore level has been applied in breeding programs with

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*Tritordeum* (= *Hordeum chilense* x *Triticum turgidum* cv. *durum*) [26], triticale [27], and with rice anther culture [28, 29]. Exposure to colchicine led to the development of diploid pollen in spikelet cultures of barley (*Hordeum vulgare* L.) and rye (*Secale cereale* L.) [30]. Colchicine-induced polyploidy has been employed widely in *Brassica* breeding and genetics [31].

But all these procedures are labour intensive [17], hazardous [8-10, 32, 33] and costly [33], because high concentrations of colchicine are needed [34]. Therefore different researchers postulated in many publications, the necessity of developing an effective alternative to colchicine [12-33].

### DINITROANILINES AND PHOSPHOROTHIOAMIDES AS ALTERNATIVE POLYPLOIDY INDUCERS

In the 1990s, attention has turned towards the potential use of chemicals other than colchicine which might further improve the *in vitro* chromosome doubling. Antimicrotubular herbicides may function as such alternatives because they have been reported to inhibit microtubule assembly in a way comparable to colchicine.

The effects of trifluralin on microtubule depolymerization and chromosome doubling in embryogenic microspore cultures of *Brassica napus* were examined and compared with those of colchicine [35]. The application of low concentrations of trifluralin was found to be a superior method for doubled haploid production. The embryos generated after trifluralin treatment developed normally, germinated readily and of the plants produced, close to 60% were fertile. The authors have demonstrated that the method based on trifluralin usage is simple, effective and inexpensive as compared with that based on colchicine.

In later studies, the potential of colchicine and the microtubule depolymerizing herbicides oryzalin, trifluralin and amiprophosmethyl (APM) for *in vitro* chromosome doubling during *B. napus* microspore culture was studied [36]. It was found that all three herbicides were similar to colchicines in effect, but at concentrations approximately 100 times lower. APM was less toxic than trifluralin and oryzalin, but no significant difference in chromosome doubling efficiency was detected between the compounds [36].

Oryzalin proved to be a more efficient chromosome-doubling agent than APM and colchicine in potato cell suspension culture [37]. Using colchicine, oryzalin and APM for the production of doubled-haploid embryos from cork oak (*Quercus suber* L.) anther culture it was found that oryzalin was more effective in inducing chromosome doubling than APM and colchicine, too [38].

The chromosome doubling capacity *in vitro* of colchicine and oryzalin in gynogenetic haploids was evaluated on *Gerbera* [39]. Summarizing the results obtained, the authors concluded that the lower dose of oryzalin may be considered superior to colchicine because of its lower phytotoxicity and the absence of long term effects and genetic disorders [39]. The effects of both of these antimitotic agents were tested also on kiwifruit, *Actinidia deliciosa*, where it was also found that trifluralin treatment in low concentration was more effective than colchicine in inducing chromosome doubling [40]. It was found recently that oryzalin is more

efficient than colchicine in inducing polyploidy of *Rhododendron* hybrids [41].

Very successful results were obtained after oryzalin-induced chromosome doubling in *Rosa* [42]. For instance, chromosome doubling by low oryzalin concentrations modified plant morphology by increases in thickness and a darker green coloration of the leaves. Internodes were longer in tetraploids than in diploids, the number of petals per flower in the tetraploid forms was double that of the diploids, and significant increases in pollen viability of chromosome-doubled hybrids were also found [42].

Four antimitotic herbicides, APM (phosphoramidate), pronamide (benzamide), oryzalin and trifluralin (dinitroanilines), were evaluated for their ability to induce chromosome doubling in anther-derived, haploid maize callus [43]. The results indicated that APM and pronamide at very low concentrations effectively induced chromosome doubling of maize anther culture-derived haploid callus. Moreover, they did not inhibit callus growth at these concentrations and the treated callus retained a high plant regeneration capacity. Thus, these two compounds could be used as alternatives to colchicine [43] for chromosome doubling.

The same herbicides were used for *in vitro* chromosome doubling in sugar and fodder beet (*Beta vulgaris*) ovule culture [44]. The best chromosome doubling results were obtained by treatment of the ovules with APM. The APM treatment showed relatively low toxicity on embryo formation in combination with a high doubling effect. Oryzalin and trifluralin had more severe toxic effects, which reduced embryo formation, thereby lower percentages of chromosome doubled plants were obtained. Pronamide had no significant toxic effect but it induced chromosome doubling at lower frequencies. Compared to colchicines, APM was found to be as efficient for chromosome doubling during beet ovule culture, but at molar concentrations 100 times lower than those used for chromosome doubling with colchicine [44].

The successful use of APM for chromosome doubling was also demonstrated on the forage plant, eastern gamagrass (*Tripsacum dactyloides* L.) [45]. The resulting tetraploid plants were morphologically normal and produced normal seeds [45]. In more recent studies, the testing of colchicine, trifluralin and oryzalin on polyploidization of anther culture of peace lily (*Spathiphyllum wallisii*) plants it was found that colchicine could be efficiently replaced by oryzalin or trifluralin which very effective in low micromolar concentrations [46]. Finally, a very broad spectrum of antimicrotubular herbicides was proposed as a colchicine alternative for efficient production of doubled haploids from microspore cultures [19].

Some time ago we reported for the first time about the comparative polyploidizing effects of colchicine, APM, and a wide spectrum of different dinitroanilines (oryzalin, ethalfluralin, trifluralin, pendimethalin, benefin), which have been assessed for production of new nep (*Nepeta* sp.) polyploid forms [47]. It was established that all dinitroanilines possessed a higher polyploidization potential than colchicine. The use of polyploidy in breeding of nep species has the potential of elevation of productivity of secondary metabolism, development of ornamental characteristics, and in

interspecific hybridization, the restoration of  $F_1$ -sterility at the tetraploid level.

### MECHANISM OF COLCHICINE-TUBULIN INTERACTION

From the various studies, it has been found that the main mechanism of colchicine action is that it blocks cell division by disrupting microtubules. After summarizing the data of different researchers it was concluded that the spindle microtubules are more sensitive to colchicine than the interphase microtubules [48]. The cells blocked by colchicine at mitosis undergo an abnormal mitotic cycle, designed as “c-mitosis” or “colchicine-mitosis”. C-mitosis is characterized by partial or complete absence of a spindle apparatus following the breakdown of the nuclear envelope, condensed chromosomes, and undivided centromeres [48, 49].

It was established that colchicine has a high affinity for soluble tubulin; however it does not bind to microtubules unless it first forms a tubulin-colchicine-complex (TC-complex), which attaches to the microtubule ends [48]. Kinetic analysis of the inhibition interaction suggests that the TC-complex binds to the microtubule ends and prevents the microtubule growth by sterically blocking further addition of the animal tubulin dimers to the ends [50].

Moreover, it was established that at a low TC-complex concentration, the complex incorporates into a microtubule by disturbing the formation of lateral contacts at the newly formed ends of the protofilaments while the microtubule still remain intact. On increasing the concentration of colchicine, a greater loss of lateral contacts leads to disassembly of microtubules [48].

Many attempts have been made by different investigators to define the exact colchicine binding site. To gain an insight into this the tubulin-colchicine crystal structure has been reported [51]. This complex threw light on the mechanism of tubulin-colchicine interaction. It was found that the A and C colchicine rings interact with  $\beta$ -subunit of tubulin and the B ring side chain interacts with the  $\alpha$ -tubulin [48-50].

Several studies have shown that colchicine has a much lower affinity for plant tubulins than for animal tubulins [52-54]. Among the four eukaryotic families (plants, animals, fungi and protists) colchicine is known to bind most strongly to animal tubulins, with an affinity constant  $\sim 10^6 M^{-1}$  [55]. Plant tubulins also bind colchicine only weakly as has been found from comparative binding assays [52]. Consequently, millimolar concentrations of colchicine are usually required to inhibit plant cell division and induce chromosome doubling, which compares unfavorably with the micromolar concentrations that affect the microtubule-associated processes in animals [56].

Recently, it was shown that a unique feature of the colchicine-binding site is its striking animal-specificity [55]. To identify this specificity two working hypotheses were considered by these authors, one concerning identification of residues responsible for animal specific colchicine binding to tubulin, and the second one related to definition of primary and extended colchicine-binding sites on tubulin molecules [55]. It was proposed that if the “collective nature” of a subset of residues in animal tubulins is distinct from that of a similar subset in non-animal tubulins, then one or more of

the residues in the animal tubulin subset is responsible for the animal specific colchicine binding [55]. Moreover, such a subset of residues must, directly or indirectly, be involved with colchicine binding as evident from crystal structure and experimental data.

It was found that twenty tubulin residues constituted the primary colchicine binding site (PBS). These residues in tubulin sequences across eukaryotes were then identified by multiple sequence alignment of numerous  $\alpha$ - and  $\beta$ -tubulin sequences from different animals, fungi, protists and plants, and it was established that only 16 of them are considered to constitute the PBS [55]. But principal component analysis has shown that the amino acid residues comprising such sites are not sufficient to reflect the animal-specific colchicine binding trait of tubulin [55].

After that the principal component analysis in the extended colchicine binding sequence space was carried out across different eukaryotes. It was demonstrated that the animal tubulin extended colchicine binding site residues are distinct – suggesting that this is sufficient to explain the origin of the animal-specificity of colchicine-tubulin interaction [55]. Two sequence positions were found as animal specific: Ala250 $\beta$  and Pro270 $\beta$ , whereas Ser250 $\beta$  and Val/Ile270 $\beta$  are characteristic for plants, fungi and protists [55]. These authors also compared the sequence conservation at binding sites of two other ligands: GTP and GDP, bound to  $\alpha$ - and  $\beta$ -tubulin, respectively. It was found that no residue positions show family-specific conservation.

Thus, from an analysis of abundance of  $\alpha$ - and  $\beta$ -tubulin sequences and colchicine-tubulin crystal structure it has been clearly demonstrated that two residues in  $\beta$ -tubulin, at sequence position Pro268 $\beta$  and Ala248 $\beta$  (270 $\beta$  and 250 $\beta$  in the crystal structure 1SA0) in animal tubulin are crucial for the observed animal-specific colchicine binding to tubulin across eukaryotes [55].

### STRUCTURAL PLANT TUBULIN PECULIARITIES RESPONSIBLE FOR EFFICIENT DINITROANILINE AND PHOSPHOROTHIOAMIDATE BINDING

In comparison with colchicine, such antimicrotubule compounds as dinitroanilines and phosphorothioamides bind more specifically than this compound to plant tubulin *in vitro* [57]. Among the low-molecular weight tubulin ligands the dinitroaniline and phosphoroamidate herbicides are compounds with a higher specificity for plant tubulins than for animal tubulins [58-61]. Earlier it was proposed that these two chemically distinct classes of antimitotic herbicides bind the same receptor sites [62].

To understand the peculiarities of dinitroaniline and phosphorothioamidate interaction with plant tubulin we have modelled the spatial structure of higher plant tubulins, based upon the established three-dimensional structure of porcine tubulin (<http://www.rcsb.org/pdb>) [63]. The reconstruction was based on the tubulin sequences of well characterized sensitive and resistant to dinitroaniline and phosphorothioamidate herbicide biotypes of goosegrass *Eleusine indica* [64]. Earlier it was shown that this resistance is caused by a point mutation at position 239 of the  $\alpha 1$ -tubulin molecule, which results in the replacement of the amino acid Thr by Ile [58-65]. It should be noted that residue Thr239 is highly

conserved in all known fungal, plant, and animal  $\alpha$ -tubulins. An analysis of the sequences flanking the mutation site showed that they also are conserved, though not identical in all plant  $\alpha$ -tubulins.

To analyze the interaction specificity of dinitroanilines and phosphorothioamides with tubulin from resistant and sensitive biotypes of *E. indica*, three-dimensional structures of both types of  $\alpha$ -tubulins were reconstructed, including the electrostatic potential distribution on their surface [64-66]. A comparison of the spatial models of  $\alpha$ -tubulins from resistant and sensitive biotypes with regard to the position of Thr239 made it possible to identify the dinitroaniline binding site. It is located on the surface of the  $\alpha$ -tubulin molecule in the interdimer contact zone [64]. It was found that this area of the surface forms a clearly shaped cavity with a highly allocated positive electrostatic potential [64].

The substitution of Leu for Thr at position 239 leads to considerable changes in the distribution of electrostatic charge on the surface of the  $\alpha$ -tubulin molecule. This redistribution of surface energy, as a result of the C to T point mutation, is apparently due to the rearrangement of side chains of neighbouring amino acid residues located on and near the surface of the tubulin molecule. This process is accompanied simultaneously by conformational changes in the predicted herbicide binding site on the  $\alpha$ -tubulin molecule. Thus the substitution of Thr239, which is located in the vicinity of the seventh helix, can markedly alter the three-dimensional structure of the  $\alpha$ -tubulin molecule [64]. So,  $\alpha$ -tubulin from the resistant biotype of goosegrass lacks the interactive cavity. Spatial reorganization of the  $\alpha$ -tubulin surface prevents ligand binding due to closing of the cavity and redistribution of electrostatic potential, when Thr239 is replaced by Ile. The comparative analysis of the surface structure of animal  $\alpha$ -tubulin shows that animal  $\alpha$ -tubulin, like plant mutated  $\alpha$ -tubulin, does not possess on its surface the cavern of interaction [64].

Modelling the Thr239Ile substitution revealed a drastic decrease in the positive electrostatic potential at the herbicide interaction site [64-66]. As a result a redistribution of charge, accompanied by partial closing (narrowing) of the interaction cavity, takes place. Therefore, the affinity of this site for dinitroaniline and phosphorothioamide herbicides appears to decrease dramatically [64-66].

Analysis of the three-dimensional structure of trifluralin and amiprofosmethyl, the most effective representatives from dinitroanilines and phosphorothioamides, respectively, revealed a similar spatial geometry for both types of molecules as well as similarly distributed surface potentials [64]. All active compounds from these herbicide groups are characterized by the presence of common structural elements – nitro groups that are linked to the benzene ring. This basic similarity implies that the polar (electronegative) moiety of their molecules should play a key role in interaction with tubulin. Therefore, we performed spatial docking of dinitroanilines and phosphorothioamides into the binding sites on the surface of  $\alpha$ -tubulin molecules from sensitive and resistant *E. indica* lines.

It was found that the hydrocarbon tails of the herbicides are oriented along amino acid residues Gly133, Asn253, and Gly256. These amino acids are located on the surface of the

$\alpha$ -tubulin molecule, and are involved in interaction with amino acid residues of the  $\beta$ -tubulin molecule of the next heterodimer. A clearly distinguishable and distinct cavity, with a high electrostatic charge, is formed in this area on the interdimer surface of the  $\alpha$ -tubulin molecule [64]. Although amino acid residue Thr239 is not exposed on the surface of  $\alpha$ -tubulin, it is located immediately under this cavity. Interactions of negatively charged  $\text{NO}_2$ -groups, which is a common structural feature of both dinitroanilines and phosphorothioamides, with the identified binding site on the  $\alpha$ -tubulin molecule can have significant consequences leading to the prevention of further microtubule polymerization. Furthermore, the position of the bound herbicide ligand will be stabilized as a result of additional interaction with negatively charged groups (such as  $-\text{CF}_3$  or  $-\text{SO}_2$ ) with the  $\text{NH}_2$ -group of Arg2 [64]. Thus, these tubulin-ligand complexes acquire a sufficient level of stability so as not to dissociate under the influence of internal molecular oscillations.

Results from comparative analyses of the three-dimensional models of  $\alpha$ -tubulins from the resistant and sensitive biotypes of *E. indica* were confirmed by a subsequent investigation of the mutant  $\alpha$ -tubulin from *Setaria viridis* [67]. In that study it was concluded that the binding site for the dinitroanilines is located immediately adjacent to the contact zone between tubulin dimers, whereas modeling of the  $\alpha$ -tubulin molecule from the protozoan *Toxoplasma gondii* assumed that dinitroaniline binding site is located in zone of lateral contacts between the microtubule protofilaments [68]. In the same time, the binding sites predicted for  $\alpha$ -tubulins of higher plants and protozoans overlap partially. Therefore, it can be concluded that the binding patterns of dinitroanilines with tubulins in the two groups of eukaryotes are slightly different, likely as a result of their phylogenetic distance.

The binding of herbicides to tubulin decreases the polymerization capability of the latter, preventing the formation of microtubules and, hence, the mitotic spindle. Subsequently, the effect of herbicides on the plant cells results in a complete loss of microtubules not only during mitosis but also in the interphase and may ultimately leads to cell death.

Therefore it can be concluded that the pattern of interaction of dinitroaniline and phosphorothioamide herbicides with the plant  $\alpha$ -tubulin molecule is unique and has no analogues in other tubulin types from higher eukaryotes.

## CONCLUSIONS

For a long period of time colchicine has been considered to be most useful as a means of producing bridge species in agronomic and floricultural crops. But postulated colchicine high toxicity for plant cells, its low binding efficiency to plant tubulin initiated screening of more potent agents for polyploidization. The present review demonstrates that such antimicrotubular compounds as dinitroanilines and phosphorothioamides could be viable alternatives to colchicine for chromosome doubling *in vitro*. Their high polyploidization potential is based on specific interaction with plant tubulin molecule. Therefore advantages of these compounds over colchicine are that dinitroanilines and phosphorothioamides are much less toxic for humans than colchicine, particularly so with the very low concentrations of the compounds needed for the plant treatment.

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