

SUPPLEMENTARY MATERIAL

Table 1. Mass Spectrometric Investigations for the Study of Gaseous Metallic and Intermetallic Species

Sample in the Effusion Cell	Temperature Range in K	Gaseous Species	Ref./Year
Li	540-830	Li, Li_2 , Li_3 , Li_4	[1]/1990
Li	627-893	Li, Li_2	[2]/2001, [3]/2001
K	350-460	K, K_2 , K_3	[1]/1990
Si	1647-1917	Si, Si_6	[4]/1996
Si	1685-?	Si, Si_2	[5]/1999
Si	1763-1993	Si, Si_7 , Si_8	[6]/2001
Ni	1270-1896	Ni	[7]/2002
Ge	1351-1744	Ge, Ge_2	[8]/2002
Ge	1351-2010	Ge, Ge_3 , Ge_4	[9]/2000
Ge	1477-1719	Ge, Ge_5 , Ge_6 , Ge_7 , Ge_8	[10]/2000
Ag	1121-1603	Ag, Ag_2	[11]/1993
Sn, Sn + Au	1421-1855	Sn, Sn_4 , Sn_5 , Sn_6 , Sn_7	[12]/2000
Tm	821-1080	Tm	[13]/1990
Yb	669-782	Yb	[13]/1990
Si + BN	2052-2146	B, Si_2N , B_2N , Si	[14]/2000
Na + Te	900-1200	Te, Te_2 , NaTe , NaTe_2 , Na_2Te , Na_2Te_2	[15]/1992
Mg + Au + Pd	1870-2333	Au, Mg, AuMg , Au_2	[16]/2003
Ga-In	1446-1642	Ga, Ga_2 , In, In_2 , GaIn	[17]/1998
Cs_2Sb_3 , CsSb , Cs_3Sb	785-818	Cs, Sb_2 , Sb_4 , Cs_2Sb , CsSb_2 , Cs_2Sb_2 , Cs_2Sb_4 , Cs_6Sb_4	[18]/1992
$\text{CsSb} + \text{Te}$	900-1200	Te, Te_2 , CsTe , CsTe_2 , Cs_2Te , Cs_2Te_2 , Cs_2Te_3	[15]/1992
$\text{M} + \text{MTe}_x$, M = Fe, Ni, Cr, Mo	820-1467	Te, Te_2	[19]/1994
$0.92\text{Pd} + 0.08\text{Pb}$	1935	Pd, Pb, PdPb	[20]/2000
$0.5\text{Pd} + 0.5\text{Sn}$	2025	Pd, Sn, PdSn	[20]/2000
$\text{Ag} + \text{Ag}_2\text{Te}$	1073-1319	Te, Te_2 , Ag, Ag_2 , AgTe , Ag_2Te	[21]/1996
$\text{Au} + \text{In}$		Au, In, Au_2 , AuIn , Au_2In , AuIn_2	[22]/1990
$\text{Au} + \text{Pb}$		Au, Pb, Pb_2 , AuPb	[22]/1990
$\text{Ge} + \text{Sn} + \text{Si}$	1704-1788	Ge, Sn, Ge_2 , Sn_2 , GeSn , Ge_2Sn , GeSn_2 , Ge_3Sn , Ge_4Sn , SnC	[23]/1998
$0.5\text{Sn} + 0.5\text{Bi}$	884-984	Bi, Bi_2 , SnBi , SnBi_2 , SnBi_3	[24]/2002, [25]/2003, [26]/2003
$\text{Ce} + \text{Rh} + \text{Ru} + \text{Os} + \text{C}$	2533-2838	C, Os, OsC , OsC_2	[27]/2001
Sc, La, Ce	Sc 1373-1573 La, Ce 1473-1573	Sc , La , Ce	[28]/2009
$\text{Pb} + \text{Sn}$ (double oven technique)	1426-1705	Sn, Pb, Sn_2 , Pb_2 , SnPb (bond energies calculated)	[29]/2005
$\text{PuCd}_2 + \text{PuCd}_4$, $\text{PuCd}_4 + \text{PuCd}_6$	525-641 490-602	Cd	[31]/2006

Table 2. Mass Spectrometric Investigations for the Determination of Thermodynamic Properties of Solid or Liquid Alloys and their Vaporisation. (Solid Compounds Present in the given Concentration Range are Underlined if the Thermodynamic Functions of Formation were Determined)

Sample in the Effusion Cell	State of Condensed Phase	Solid Compounds	Temperature Range in K	Gaseous Species	Ref./Year
xLi + (1-x)Al, 0.017 < x < 0.298	s	<u>fcc</u>	665-918	Li	[32]/2002
0.25Li + 0.75Sn	l		800-1200	Li, SnLi	[33]/2002
xMg + (1-x)Cu, x = 0.1-0.8	s	<u>Cu₂Mg, CuMg₂</u>	590-860	Mg	[34]/1998
xMg + (1-x)Cu, 0.027 < x < 0.870	l		873-1073	Mg, Cu	[35]/2000
xCa + (1-x)Cu, 0.024 < x < 0.888	l		973-1173	Ca, Cu	[35]/2000
xAl + (1-x)Ti, x = 0.45, 0.62	s	<u>TiAl</u>	1223-1674	Al, Ti	[36]/1999
xAl + (1-x)Ti, 0 < x < 1	s			Ti, Al	[37]/1997
x ₁ Al + x ₂ Ti + (1-x ₁ -x ₂)Nb	s	<u>Ti₃Al, TiAl</u>	1170-1635	Ti, Al	[38]/1999 [37]/1997
x ₁ Al + x ₂ Ti + (1-x ₁ -x ₂)Cr	s		1228-1521	Al, Ti, Cr	[36]/1999
xAl + (1-x)Mn, 0.58 > x > 0.38	l		1250-1550	Al, Mn	[39]/1994
xAl + (1-x)Mn, 0.58 > x > 0.38	s	<u>Al_{0.57}Mn_{0.43}, Al_{0.50}Mn_{0.50}, Al_{0.45}Mn_{0.55}</u>	1100-1275	Al, Mn	[40]/1992
xAl + (1-x)Fe x = 0.815, 0.659, 0.526, 0.420	l		1410	Al	[41]/1996
xAl + (1-x)Fe, 0.656 > x > 0.097	s, l		1500-1670	Fe, Al	[42]/1993
xAl + (1-x)Cu, 0.955 > x > 0.091	l		1150-1680	Al, Cu	[43]/1998
xAl + (1-x)Ni, 0.3 > x > 0	s, l	Ni ₃ Al	1389-1734	Ni, Al	[44]/1990, [37]/1997
(Al _{0.22} Ni _{0.78}) _{0.995} B _{0.005}	s	Ni ₃ Al	1400-1670	Ni, Al	[45]/1990
x(Ni ₃ Al) + (1-x)Fe, 0 < x < 0.08	s	Ni ₃ Al	1304-1698	Ni, Al, Fe	[46]/1996 [37]/1997
x(Ni ₃ Al) + (1-x)Co, 0 < x < 0.15	s	Ni ₃ Al	1326-1581	Ni, Al, Co	[47]/1997 [37]/1997
(0.25-x)Al + 0.75Ni + xHf; x(Ni ₃ Al) + (1-x)Hf, 0 < x < 0.10	s	Ni ₃ Al	1361-1680	Ni, Al	[48]/1992 [37]/1997
(0.25-x)Al + 0.75Ni + xTi, 0 < x < 0.25	s	Ni ₃ Ti	1350-1650	Ni, Al, Ti	[49]/1996
Ti _{0.58} Ge _{0.42}	s	<u>Ti₅Ge₃, Ti₆Ge₃</u>	1665-1880	Ti, Ge	[50]/2000
xV + (1-x)Ge, 0.6 < x < 0.9	s	<u>V₃Ge₁₋₃, V₅Ge₃, V₁₁Ge₈, V₁₇Ge₃₁</u>	1750-1980	V, Ge	[51]/2000
xCr + (1-x)Ge, 0.625 < x < 0.93	s	Cr ₃ Ge, Cr ₅ Ge ₃	1400-1600	Cr, Ge	[52]/2001
xCr + (1-x)Fe, 0.906 > x > 0.012	s			Cr	[53]/1990
xCr + (1-x)Fe, 0 < x < 1	s		1304-1630	Cr	[53]/1990 [54]/1990
xCr + (1-x)Ni, 0.6989 < x < 0.9405	s	bcc	1300-1650	Cr, Ni	[55]/1995
xCr + (1-x)Ni, 0.102 < x < 0.636	l		1773	Cr, Ni	[56]/1993
x(Ni _{0.86} Cr _{0.14}) + (1-x)Fe	s	fcc	1673	Fe, Ni, Cr	[57]/1995
x ₁ Cr + x ₂ Ni + (1-x ₁ -x ₂)Fe	s	fcc	1650	Fe, Cr, Ni	[58]/1998
x ₁ Cr + x ₂ Ni + (1-x ₁ -x ₂)Fe	l		1850-1950	Fe, Cr, Ni	[59]/1998 [58]/1998
xCr + (1-x)Co, 0.145 < x < 0.9227	s	bcc, fcc	1580-1790	Cr, Co	[60]/1998
xCr + (1-x)Co, 0.272 < x < 0.605	l		1710-1860	Cr, Co	[61]/1999
xCr + (1-x)Co, 0.050 < x < 0.501	l		1773	Cr, Co	[56]/1993

Table 2. Contd...

Sample in the Effusion Cell	State of Condensed Phase	Solid Compounds	Temperature Range in K	Gaseous Species	Ref./Year
$x\text{Mn} + (1-x)\text{Fe}$, $0 < x < 1$	s		967-1410	Mn	[54]/1990
$x\text{Mn} + (1-x)\text{Cu}$, $0.7314 > x > 0.0462$	l		1310-1480	Cu, Mn	[62]/1996
$x\text{Mn} + (1-x)\text{Te}$, $0 < x < 0.5$	s	<u>MnTe</u> , <u>MnTe₂</u>	650-750 K	Te ₂	[63]/2002
$x\text{Fe} + (1-x)\text{Cu}$, $0.899 < x < 0.9983$	l		1770-1916	Fe, Cu	[64]/2001
Fe_2Zr	s	Fe_2Zr	1550-1756	Fe	[65]/1993
$x\text{Fe} + x\text{Ni} + (1-2x)\text{Co}$	s		1500-1690	Co, Fe	[66]/1993
$x\text{Fe} + x\text{Ni} + (1-2x)\text{Co}$	l		1850	Co, Fe	[67]/1993
$x_1\text{Fe} + (1-x_2)\text{Ni} + (1-x_1-x_2)\text{Cu}$	l		1623	Ni, Cu, Fe	[68]/1997
$x\text{Co} + (1-x)\text{Ni}$, $0.096 < x < 0.887$	l		1773	Co, Ni	[56]/1993
$x\text{Ni} + (1-x)\text{Zr}$, $0.012 < x < 0.994$	s	Ni_5Zr , Ni_7Zr_2 , $\text{Ni}_{21}\text{Zr}_8$, $\text{Ni}_{10}\text{Zr}_7$, $\text{Ni}_{11}\text{Zr}_9$, NiZr , NiZr_2	971-1859	ZrF ₄ , Na, Mg, Ca	[69]/2003 [70]/2002
$x\text{Ni} + (1-x)\text{Cu}$, $0 < x < 0.5$	l		1623	Ni, Cu	[68]/1997
$x\text{Ni} + (1-x)\text{Yb}$	s	$\text{Ni}_{17}\text{Yb}_2$, Ni_5Yb , Ni_3Yb , Ni_2Yb , NiYb	734-1444	Yb	[71]/2001
$x\text{Cu} + (1-x)\text{Zr}$, $0.027 < x < 0.949$	s	Cu_9Zr_2 , $\text{Cu}_{51}\text{Zr}_{14}$, Cu_8Zr_3 , Cu_2Zr , $\text{Cu}_{10}\text{Zr}_7$, CuZr , Cu_5Zr_8 , CuZr_2	701-1740	ZrF ₄ , K, Na, Mg, Ca	[72]/2003
$x_1\text{Cu} + x_2\text{Au} + (1-x_1-x_2)\text{Ge}$	l		1101-1762	Cu, Au, Ge	[73]/1994
$x\text{Cu} + (1-x)\text{Ag}$ $x = 0.850, 0.627, 0.390, 0.211$	l		1310	Ag	[41]/1996
$x_1\text{Cu} + x_2\text{Ag} + (1-x_1-x_2)\text{Ge}$	l		1416	Ag	[74]/1994
$x\text{Cu} + (1-x)\text{Gd}$, $0.156 < x < 0.724$	l	<u>CuGd</u> , <u>Cu₂Gd</u>	1282-1664	Cu	[75]/1995
$x\text{Cu} + (1-x)\text{Nd}$, $0 < x < 1$	l		1600	Cu, Nd	[76]/1993
$x\text{ZnTe} + (1-x)\text{CdTe}$, $0 < x < 1$	s	s.s.	900	Zn, Cd, Te ₂	[77]/2002 [78]/2003 [79]/2003
$\text{GeTe} + \text{Ge}$	s	GeTe , Ge	700-850 K	GeTe , Te_2 , GeTe_2	[80]/2002
Cr_2Zr	s	Cr_2Zr	1476-1704	Cr	[65]/1993
$x\text{Te} + (1-x)\text{Cr}$, $0.60 < x < 0.827$	s	<u>CrTe_{4+y}</u> , <u>CrTe₃</u> , <u>Cr₅Te₈</u>	650-1120	Te ₂	[81]/1994 [82]/1996
$x\text{Co} + (1-x)\text{Cu}$, $0.141 < x < 0.750$	l		1347-1587	Co, Cu	[83]/2000
In_2Te_3 , InTe	s	In_2Te_3 , InTe	443-667	In, Te, Te ₂ , InTe , In_2Te , InTe_2 , In_2Te_2	[84]/2000
$x\text{Te} + (1-x)\text{Mn}$, $x = 0.496, 0.549$	s	MnTe	1194-1343	Mn, Te, Te ₂	[85]/1998 [82]/1996
$x\text{Te} + (1-x)\text{Au}$, $0.48 < x < 0.67$	l		753	Te ₂	[86]/1997
$\text{Mo}_3\text{Te}_4 + \text{MoTe}_2$	s	<u>Mo₃Te₄</u> , MoTe_2	820-950	Te, Te ₂	[87]/1993 [82]/1996
$x\text{Hf} + (1-x)\text{Al}$, $0.25 < x < 0.50$	s	<u>HfAl</u> , <u>HfAl₃</u> , <u>Hf₂Al₃</u> , <u>Hf₅Al₄</u>	1280-1680	Al	[88]/1995
$x\text{Hf} + (1-x)\text{Ni}$, $0.0501 < x < 0.9408$	s	<u>NiHf₂</u> , <u>NiHf</u> , <u>Ni₁₁Hf₆</u> , <u>Ni₁₀Hf₇</u> , <u>Ni₇Hf₄</u> , <u>Ni₂₇Hf₈</u> , <u>Ni₃Hf</u> , <u>Ni₇Hf₂</u> , <u>Ni₃Hf</u>	1423-1780	Ni	[89]/1996
$\text{Pd}_{1-x}\text{Zr}_x$, $x = 0.04, 0.08, 0.12$ $\text{Pd} + \text{Pd}_3\text{Zr}$, $\text{Pd}_5\text{Zr} + \text{Pd}_2\text{Zr}$	s	<u>Pd₃Zr</u> , <u>Pd₂Zr</u>	1518-1981	Pd	[90]/1990

Table 2. contd...

Sample in the Effusion Cell	State of Condensed Phase	Solid Compounds	Temperature Range in K	Gaseous Species	Ref./Year
xPd + (1-x)Nb, 0.60 < x < 0.97	s		1450-1750	Pd	[91]/1991
Ru + RuGe; RuGe + Ru ₂ Ge ₃	s	RuGe, Ru ₂ Ge ₃	1529-1760	Ge	[92]/1998
Ru ₃ Sn ₇ + Ru ₂ Sn ₃ ; Ru ₂ Sn ₃ + Ru	s	Ru ₃ Sn ₇ , Ru ₂ Sn ₃	1317-1523	Sn	[92]/1998
Pd ₃ In + Pd	s	Pd ₃ In, Pd _{0.83} In _{0.17}	1382-1495	Pd, In	[20]/2000
Pd ₃ Pb + Pd	s	Pd ₃ Pb, Pd _{0.85} Pb _{0.15}	1191-1438	Pd, Pb	[20]/2000
Pd ₇ Sn ₂ + Pd ₃ Sn; Pd ₇ Sn ₂ + Pd	s	Pd ₃ Sn, Pd ₇ Sn ₂ , Pd _{0.85} Sn _{0.15}	1400-1553	Pd	[20]/2000
xPd + (1-x)Ni, 0.2005 < x < 0.8003	l		1850	Pd, Ni	[93]/1993
xPd + (1-x)Ni, 0.2005 < x < 0.8003	s		1500	Pd, Ni	[94]/1992
xPd + (1-x)Co, 0.10 < x < 0.80	s	fcc	1470	Pd, Co	[95]/1995
xPd + (1-x)Au, 0.1015 < x < 0.8953	l		1850	Pd, Au	[96]/1990
xPd + (1-x)Au, 0.2921 < x < 0.8953	s		1500	Pd, Au	[97]/1990
x ₁ Pd + x ₂ Pt + (1-x ₁ -x ₂)Ge	l		1820	Pd, Ge	[98]/1999
x ₁ Ag + x ₂ In + (1-x ₁ -x ₂)Sn	l		1273-1523	Ag, In	[99]/2001
xLa + (1-x)Bi	s	LaBi ₂ , LaBi, La ₄ Bi ₃ , La ₅ Bi ₃ , La ₂ Bi	795-1066	Bi ₂	[100]/2000
xLa + (1-x)Ce, 0 < x < 1	l		1592-1781	La, Ce	[101]/1998
xLa + (1-x)Nd, 0 < x < 1	l		1474-1767	Nd	[102]/1998
xLa + (1-x)Gd, 0 < x < 1	l		1588-1797	La, Gd	[103]/1997
xLa + (1-x)U, 0.01 < x < 0.99	l		1619-1886	La, U	[104]/1999
xCe + (1-x)U, 0.01 < x < 0.99	l		1619-1886	Ce, U	[104]/1999
xFe + (1-x)U, 0.05 < x < 0.90	s,l	UFe ₂ , U ₆ Fe	1440-1850	Fe	[105]/1992
0.997U + 0.003Fe	l		1780-2180	U, Fe	[106]/1993
xGa + (1-x)U, 0.05 < x < 0.28	s,l	U ₂ Ga ₃	1100-1670	Ga	[105]/1992
xZr + (1-x)U, x = 0.244, 0.393	l		1700-1973	U	[107]/1992
xZr + (1-x)Pu, 0.20 < x < 0.94	s, l		1400-1900	Pu	[108]/1994
Aluchrom (Ni/Cr/Al base)	s		1313-1556	Ni, Cr, Al	[109]/1992
ODS MA 6000 (Fe/Cr/Al base)	s		1393-1562	Fe, Cr, Al	[109]/1992
M5-alloy (71Fe20Ni6Cr3Mn)	s		1550-1900	Fe, Cr, Ni	[110]/1993
steel AISI 316 (Fe, Ni, Cr base)	s, l		1293-2120	Fe, Cr, Ni, Mn	[111]/1995
Zircaloy-4 (Zr-base) + Te	s, l		1062-2315	Te ₂ , SnTe, Sn, Fe, Cr	[112]/1990
xNi + (1-x)La 0 < x < 1 over the "entire compositions range"	l		1023-1854	Ni, La	[113]/2005
Al + Mn (0-50.1 at % Mn)	l		1043-1670	Al, Mn	[114]/2006
Al+Fe+Ni, B2	s		1180-1508	Al, Fe, Ni,	[115]/2006
xAl+(1-x)Cu 0 < x < 1	l		1035-1727	Al, Cu	[116]/2007
xNi+(1-x)La 0 < x < 1	s, l		685-1854	Ni, La	[117]/2007
Haynes 230, Inconel 617 Ni-based alloy, commercial alloys	s		1423-1548	Cr	[118]/2008
(1-x)Al+xMn 0 < x < 50.1 (1-x)Al+xCu, (1-x)Al+xFe, 0 < x < 1 (1-x ₁ -x ₂) Al+x ₁ Fe+x ₂ Cu 0 < x ₁ , x ₂ < 1	l		1043-1670 1123-1878	Cu, Mn	[119]/2008
(1-x)Ni+xSb 0.20 < x < 0.54	s	Ni ₃ Sb, γ-NiSb	773-1423	Sb, Sb ₂ , Sb ₃ , Sb ₄	[120]/2007

Table 2. contd...

Sample in the Effusion Cell	State of Condensed Phase	Solid Compounds	Temperature Range in K	Gaseous Species	Ref./Year
$(1-x)\text{Yb}+x\text{Ge}$ $0.35 < x < 1$	s	Yb_2Ge , Yb_5Ge_3 , Yb_5Ge_4 , $\text{Yb}_{11}\text{Ge}_{10}$, Yb_3Ge_5	801-1455	Yb, Ge	[121]/2007
$(1-x)\text{Fe}+x\text{Zr}$ $x=50-85$ at%	l		1503-1843	Fe	[122]/2005
$(1-x_1-x_2)\text{Cu}+x_1\text{In}+x_2\text{Sn}$, $x_1=0.20, 0.25, 0.29, 0.30, 0.40$ $\text{Sn}/\text{Cu}=1$ a $0.05 < x_1 < 0.8$	l		1273-1473	Cu^+/Sn^+	[123]/2006
$(1-x)\text{Ni}+x\text{La}$ $0 < x < 1$	s, l	NiLa , Ni_2La , Ni_3La , $\text{Ni}_{17}\text{La}_4$, $\beta\text{-Ni}_7\text{La}_2$, fcc Ni , fcc La	685-1854	Ni, La	[124]/2006
$(1-x)\text{Al}+x\text{Mn}$ $x=0-26$ at% $x=0-50.1$ at%	s l	$\text{fccAl}+a\text{Mn}$	628-1193 1043-1670	Mn, Al	[125]/2006
$(1-x_1-x_2)\text{Ag}+x_1\text{In}+x_2\text{Sn}$, $x_1=0.10, 0.117, 0.20, 0.30, 0.40, 0.50$	l		1273-1473	Ag^+/Sn^+ , Ag^+/In^+	[126]/2008
$(1-x)\text{Al}+x\text{Mn}$ $0 < x < 0.26$ $0 < x < 0.501$	s, l	$a\text{Mn-fcc-Al}$	628-1193 1043-1670	Mn, Al	[127]/2008
$(1-x)\text{Fe}+x\text{Zr}$ $x=50-90$ at%	l, l+s	Fe_2Zr	1501-1852	Fe, Zr	[128]/2008
3 different commercial 18ct. gold alloys	s, l		773-2273	Au, Ag, Cu	[129]/2008
45.1Fe+39.8Al+15.1Cr (at%) 49.9Fe+19.9Al+30.2Cr (at%)	l, s		1431-1853	Fe, Al, Cr	[130]/2008
$(1-x_1-x_2)\text{Fe}+x_1\text{Co}+x_2\text{Cr}$ $x_2/x_1=9$ and 3	s	bcc	1673	Fe, Cr	[131]/2009
$(1-x_1-x_2)\text{Fe}+x_1\text{Co}+x_2\text{Cr}$ $x_2/x_1=2/3$ and $x_2/x_1=1/4$	s	fcc	1673	Fe, Cr	[132]/2009
$(1-x_1-x_2)\text{Au}+x_1\text{Co}+x_2\text{Pd}$ $x_1/(1-x_1-x_2)=1$ and $x_2/x_1=1$	l		1800	Au, Pd	[133]/2009
$\text{Au(s)} + \text{H}_2\text{(g)}$ $\text{Ga(s)} + \text{H}_2\text{(g)}$	l		1542-1678 1258-1260	Au, H ₂ , AuH Ga, Ga ₂ , GaH	[134]/2007
$x\text{Ni}+(1-x)\text{Al}$ $0.32 < x < 0.50$	s, l	$\text{Ni}_{0.5}\text{Al}_{0.5}$, $\text{Ni}_x\text{Al}_{(1-x)}$ $0.32 < x < 0.50$, Al_3Ni_2 , AlNi	1077-1592	Al, Ni	[135]/2006
$(1-x)\text{Al}+x\text{Ni}$, $x=0.43, 0.42, 0.35, 0.32$	s	$\text{Al}_3\text{Ni}_{1.747}$, $\text{Al}_3\text{Ni}_{2.025}$, $\text{AlNi}_{0.766}$, AlNi_3	1104-1278 1021-1122	Al	[136]/2006

Table 3. Mass Spectrometric Investigations for the Determination of Thermodynamic Properties of Solid or Liquid Systems Metal-Nonmetal. (Solid Compounds are given if the Thermodynamic Functions of Formation were Determined)

Sample in the Effusion Cell	State of Condensed Phase	Solid Compounds	Temperature Range in K	Gaseous Species	Ref./Year
$x\text{B} + (1-x)\text{Si}$, $0.015 < x < 1$	s, l	SiB_6 , SiB_{14} , SiB_{15} , SiB_{40} , SiB_n , B(ss)	1522-1880	B, Si, Si ₂	[137]/2000
MgB_2 , MgB_4	s	MgB_2 , MgB_4 , MgB_7	883-1154	Mg	[138]/2002
$x\text{Si} + (1-x)\text{P}$, $0.735 < x < 0.991$	l		1507-1831	P, P ₂	[139]/2001
$x\text{Ca} + (1-x)\text{Si}$, $0.25 < x < 0.80$	s	Ca_2Si , Ca_5Si_3 , CaSi , $\text{Ca}_{14}\text{Si}_{19}$, CaSi_2	971-1341	Ca	[140]/2001
$x\text{Ti} + (1-x)\text{Si}$	l		2733	Si, Ti	[141]/2000
$x\text{Cr} + (1-x)\text{P}$, $0.75 < x < 0.87$	s, l	Cr_3P , Cr_{12}P_7	1341-1819	Cr, P, P ₂	[142]/1998
$x_1\text{Cr} + x_2\text{Fe} + (1-x_1-x_2)\text{P}$	l		1403-1821	Fe, Cr, P, P ₂	[143]/1998 [144]/1998
$x_1\text{Mn} + x_2\text{Fe} + (1-x_1-x_2)\text{P}$	l		1302-1701	Fe, Mn, P, P ₂	[143]/1998
$x_1\text{Mn} + x_2\text{Fe} + (1-x_1-x_2)\text{Si}$	l		1435-1809	Fe, Mn, Si, Si ₂	[143]/1998

Table 3. contd...

Sample in the Effusion Cell	State of Condensed Phase	Solid Compounds	Temperature Range in K	Gaseous Species	Ref./Year
$x\text{Fe} + (1-x)\text{B}$, $0.270 < x < 0.903$	l		1473-1854	Fe, B	[145]/2002
$x\text{Fe} + (1-x)\text{B}$, $0.270 < x < 0.903$	s	FeB, Fe ₂ B	1449-1854	Fe, B	[146]/2001 [147]/2003
$x\text{Fe} + (1-x)\text{As}$, $0.8999 < x < 0.9901$	l		1624-2013	Fe, As, As ₂	[148]/1991
Fe + Cu + S + As	s,l		1473	As, As ₂ , AsS	[149]/2001
$x\text{Ni} + (1-x)\text{P}$, $0.675 < x < 0.74$	s	Ni ₃ P, Ni ₅ P ₂ , Ni ₁₂ P ₅ , Ni ₂ P	971-1440	Ni, P ₂	[150]/2003
$x_1\text{Cu} + x_2\text{Ni} + (1-x_1-x_2)\text{S}$	l		1500	Cu, S ₂	[151]/2000
$x_1\text{Cu} + x_2\text{Ag} + (1-x_1-x_2)\text{Cu}_2\text{S}$	l		1373-1473	Ag	[152]/1998
$x_1\text{Cu} + x_2\text{Au} + (1-x_1-x_2)\text{Si}$	l		1750	Si, Cu, Au	[153]/1998
$x\text{Ge} + (1-x)\text{Si}$, $0.087 < x < 0.921$	l		1723	Si, Ge	[154]/1992
$x\text{Yb} + (1-x)\text{Si}$	s	Yb ₅ Si ₃ , Yb ₅ Si ₄ , YbSi, Yb ₃ Si ₄ , YbSi _{2-x}	781-1395	Yb	[155]/2003
Ru ₂ Si ₃ + RuSi	s	Ru ₂ Si ₃	1698-1909	Si	[92]/1998
YbSi _{1.74} -liquid solution at%Si ~ 65%	l		1421-1512	Yb	[156]/2005
$(1-x)\text{Cr} + x\text{Cr}_3\text{P}$, $0 < x < 1$	s	<u>Cr₃P</u>	1523-1623	P ₂	[157]/2008
$(1-x)\text{Cu} + x\text{Cu}_3\text{P}$, $0 < x < 1$	s	<u>Cu₃P</u> , fcc	1160-1250	P ₂ , Cu	[158]/2009
MgB ₂ , MgB ₄ , MgB ₇ , MgB ₂₀	s	<u>MgB₂₀</u> , <u>MgB₇</u> , <u>MgB₂</u> , <u>MgB₄</u>	1112-1467	Mg	[159]/2005
$(1-x_1-x_2)\text{Cu}_2\text{S} + x_1\text{FeS} + x_2\text{As}$, $(1-x_1-x_2)/x_1 > 15.4$ 30 % < Cu in FeS-Cu ₂ S < 80%	l		1473	As	[160]/2005
$(1-x)\text{Sr} + x\text{Si}$ $0.34 < x < 0.78$ at%Si	s	<u>Sr₂Si</u> , <u>Sr₃Si₃</u> , <u>SrSi</u> , <u>SrSi₂</u>	665-1300	Sr	[161]/2006
$x_1\text{Fe} + x_2\text{F} + \text{C}$ $0.2 < x_1 < 0.92$ mass % 4.24 % < x ₂ < 4.40 %	l		1473-1673	P ₂	[162]/2007
$(1-x)\text{Ba} + x\text{Si}$ $0.35 < x < 95$ at%Si	s,l	<u>Ba₂Si</u> , <u>Ba₅Si₃</u> , <u>BaSi</u> , <u>Ba₃Si₄</u> , <u>BaSi₂</u>	893-1325	Ba	[163]/2008
$(1-x)\text{Si} + x\text{Sn}$ $0 < x < 1$	gas		1474-1944	SiSn, Si ₂ Sn, SiSn ₂ , Si ₃ Sn, Si ₂ Sn ₂ , SiSn ₃ , Si, Sn, Si ₂ , Sn ₂ , Si ₃ , Sn ₃	[164]/2009

Table 4. Summary of Mass Spectrometric Studies of Oxide Systems. (The Gaseous Species or Solid Compounds are Underlined if their Thermodynamic Functions of Formation or of Dissociation are given)

Sample in the Effusion Cell	Temperature Range in K	Gaseous Species	Ref./Year
Na ₂ O ₂ , Na ₂ O	1100-1162	Na, O ₂ , Na ₂ O, NaO	[165]/1993
MgO	-	Mg, O, O ₂	[166]/2000
MgO + MgCr ₂ O ₄	1750 K-2080	Cr, CrO, CrO ₂ , Mg, LaO	[167]/2001
BaO + GeO ₂	1670-1788	BaO, GeO, <u>BaGeO₂</u>	[168]/1996
Ti ₁₀ O ₁₉	1753-1946	TiO, TiO ₂ , O, O ₂	[169]/1990
TeO ₂	750-950	Te ₂ , (TeO) _x , <u>(TeO₂)_x</u> , x = 1, 2, 3	[170]/1997
V ₂ O ₅ + H ₂ O		V ₄ O ₁₀ , V ₄ O ₈ , V ₂ O ₃ (OH) ₄ , H ₂ O, O ₂	[171]/1998
ZnO	1472-1623	Zn, O ₂	[172]/1993
ZnO	450-1446	Zn, O ₂ , <u>(ZnO)</u>	[173]/2003 [174]/2002
CuO + Cu ₂ O	679-1033	O ₂	[175]/1995

Table 4. contd...

Sample in the Effusion Cell	Temperature Range in K	Gaseous Species	Ref./Year
NiO + CuO	1472-1623	Cu, CuO, Ni, NiO, O ₂	[172]/1993
ZnO + CuO	1472-1623	Zn, Cu, CuO, O ₂	[172]/1993
Pd(s) + O ₂	1764-1814	Pd, O, <u>PdO</u>	[176]/2000
Tc _x O _y + O ₂ + H ₂ O	873-1373	Tc ₂ O ₇ , TcO ₃ , TcO ₃ (OH), Tc ₂ O ₅ , TcO ₂ (OH) ₃	[177]/1993
SnO ₂	1160-1450	SnO, <u>Sn₂O₃</u> , O ₂	[178]/1999
Sn(l) + SnO ₂ (s)	820-1160	SnO, Sn ₂ O ₂ , Sn ₄ O ₄ , Sn ₆ O ₆	[179]/1996
CeO _{1.99} , Ce ₂ O _{2.96}	1700-2200	CoO, CeO ₂ , O	[180]/2000
PbO Pb(l) + O ₂	838-1018	PbO, Pb ₂ O ₂ , Pb ₄ O ₄	[181]/1994
PbO	890-1100	PbO, <u>Pb₂O₂</u> , <u>Pb₄O₄</u> , Pb, O ₂	[182]/1997
Bi ₂ O ₃	991-1015	Bi, O ₂	[183]/1998
UO _{2+x}	1473-1873	<u>UO₃</u> , O ₂	[184]/2003
UO _{2+x} + U	2000-2250	<u>U</u> , <u>UO</u> , <u>UO₂</u>	[185]/2002
UO _{2+x} + MgO	1723-1923	<u>UO₃</u> , Mg, O ₂	[186]/2001
Eu ₂ O ₃ + UO ₂ + WO ₃ EuPO ₄ + UO ₂	2150-2420	<u>UO₃</u> , <u>U₂O₂</u> , <u>U₂O₃</u> , <u>U₂O₄</u>	[187]/1991
PuO ₂ ; (U,Pu)O ₂	1800-1900	PuO ₃	[188]/2000
Li ₂ TiO ₃	1373-1573	Li, LiOH, Li ₂ O, LiO, H ₂ O, H ₂	[189]/1997 [190]/2000
Li ₂ TiO ₃ + H ₂ (D ₂)	1423-1573	Li, O ₂ , D ₂ , D ₂ O, LiOD	[191]/2003
Li ₄ TiP ₂ O ₉	1100-1300	LiPO ₃ , LiPO ₂ , PO, O ₂ , P _n , n = 1-4	[192]/1997
Li _{0.2} VO _{2.6}	730-1016	<u>LiVO₃</u> , O ₂	[193]/1996
Li ₂ ZrO ₃	1234-1689	Li, LiO, Li ₂ O, O ₂	[194]/1993 [190]/2000
Li ₈ ZrO ₆	1181-1429	Li, LiO, Li ₂ O, Li ₂ O ₂ , Li ₃ O, O ₂	[195]/1996
Li ₂ MoO ₄	1403-1504	<u>Li₂MoO₄</u> , Li ₂ O, MoO ₃	[196]/1999
Li ₂ SnO ₃	1455-1590	Li, O ₂ , LiO, Li ₂ O, SnO	[197]/1991
NaPO ₃	1044-1434	NaPO ₃ , Na ₂ P ₂ O ₆ , Na	[193]/1993
Na ₃ PO ₄	1261-1693	Na, O ₂ , NaPO ₃	[193]/1993
NaBiO ₂	1200-1250	BiO, Bi, <u>NaBiO₂</u>	[198]/1999
NaUO ₃ +Na ₂ UO ₄ +Na ₂ U ₂ O ₇	1117-1290	Na	[199]/1994
KVO ₃ , l	1160-1710	KVO ₃ , (KVO ₃) ₂	[200]/1990
K ₂ MoO ₄	1170-1320	K ₂ MoO ₄	[201]/2002
K ₂ WO ₄	1190-1300	K ₂ WO ₄	[202]/2002
KTcO ₄ + KReO ₄	< 1173	KTcO ₄ , (KTcO ₄) ₂	[203]/1993
Rb ₂ MoO ₄	1102-1307	Rb ₂ MoO ₄	[204]/2000
Cs ₂ MoO ₄	1135-1245	Cs ₂ MoO ₄	[205]/1993
Cs ₂ TeO ₄ , Cs ₂ TeO ₃	1002-1077	Cs ₂ TeO ₃ , O ₂	[206]/1994
CsOH + TcO _x + O ₂	< 1173	CsTcO ₄ , (CsTcO ₄) ₂	[203]/1993
Li ₂ WO ₄	1320-1480	Li ₂ WO ₄ , Li ₂ O, WO ₃	[207]/2000
Na ₂ WO ₄	1230-1380	Na ₂ WO ₄	[208]/2001
Cs ₂ WO ₄	1080-1200	Cs ₂ WO ₄	[209]/2001

Table 4. contd...

Sample in the Effusion Cell	Temperature Range in K	Gaseous Species	Ref./Year
Cs ₂ WO ₄	1017-1233	Cs ₂ WO ₄	[210]/2001
CsBiO ₂	1130-1150	Bi, BiO, <u>CsBiO₂</u>	[198]/1999
Cs ₂ UO ₄ + H ₂ + O ₂	873-1373	Cs, CsOH, CsO, H ₂ O, O ₂	[211]/1998 [212]/1999
Cs ₂ U ₄ O ₁₂ + H ₂ + O ₂	1273-1573	Cs, CsOH, UO ₃ , H ₂ O, O ₂	[213]/1999 [212]/1999
BeO-MoO ₃	2450-2720	(BeO) _n , n=1-5, MoO _n , n=1-3, BeMoO ₃	[214]/2001 [215]/2001
Be-TiO ₂ , BeO-Ta ₂ O ₅ , BeO-Nb ₂ O ₅	2450-2720	(BeO) _n , n=2-4, XO, XO ₂ , BeXO ₃ , X=Ti,Ta,Nb	[214]/2001
BeO-Nb ₂ O ₅	2230-2450	(BeO) _n , n=1-4, NbO, NbO ₂ , BeNbO ₃	[216]/2001
MO-Ta ₂ O ₅ , M = Be, Ca, Sr, Ba	2450-2720	(BeO) _n , n=1-4, M, MO, TaO, TaO ₂ , MTaO ₂ , MTaO ₃	[217]/1999
BeO-WO ₃	2271-2475	Be ₂ O ₃ , Be ₄ O ₄ , WO ₂ , WO ₃ , BeWO ₃ , BeWO ₄ , Be ₂ WO ₄	[215]/2001
BeO-WO ₃ , BeTaO ₃ , BeNbO ₃ , BeTiO ₃	2450-2720	(BeO) _n , n = 2-5, WO ₂ , WO ₃ , W ₂ O ₆ , W ₃ O ₉ , BeWO ₃ , BeWO ₄ , Be ₂ WO ₄ , Be ₂ WO ₅	[214]/2001
xCaO + (1-x)Al ₂ O ₃ , 0.143 < x < 0.860	1833-2033	Ca, Al, AlO, CaAlO, O ₂ , O	[218]/1995 [219]/1995
CaO-Nb ₂ O ₅ , SrO-Nb ₂ O ₅	2200-2400	MO, NbO, NbO ₂ , MNbO ₃ , M = Ca, Sr	[216]/2001
<u>SrUO₃</u> + UO ₂	1534-1917	Sr, UO ₂ , SrO	[220]/1997
<u>SrUO₃</u> + H ₂ + H ₂ O	1470-1920	Sr, H ₂ , H ₂ O	[221]/1997
<u>SrPuO₃</u> + PuO ₂	1433-1913	Sr, PuO	[222]/1997
xSrO + (1-x)BaO, 0.101 < x < 0.872	1430-1530	Sr, Ba	[223]/1992
xBaO + (1-x)Y ₂ O ₃ , x = 0.8, 0.57, 0.4	1450-1850	BaO, Ba ₂ O ₂ , Ba, O, O ₂	[224]/1995
<u>BaZrO₃</u> + ZrO ₂ ;	1899-2085	BaO	[225]/1994
BaZrO ₃ + Ba ₅ ZrO ₄	1566-1732	Ba, BaO	[225]/1994
BaO-Nb ₂ O ₅	1989-2493	Ba, BaO, NbO, NbO ₂ , BaNbO ₂ , BaNbO ₃ , BaNb ₂ O ₆	[216]/2001
Ba ₂ CaWO ₆	1620-1860	Ba, Ca, BaO, WO ₃ , <u>BaWO₄</u>	[226]/1993
Ba ₂ SrWO ₆	1620-1860	Ba, Sr, BaO, WO ₃ , <u>BaWO₄</u>	[226]/1993
<u>BaUO₃</u>	1770-1920	BaO, Ba, UO ₂	[227]/1996
BaUO ₃ + H ₂ + H ₂ O	1470-1920	BaO, H ₂ O, H ₂	[221]/1997
BaPuO ₃	1673-1873	Ba, BaO	[228]/1999
BPO ₄	1447-1540	B ₂ O ₃ , P ₄ O ₁₀ , PO, PO ₂ , <u>BPO₄</u> , <u>BPO₃</u>	[229]/1999 [230]/1998 [231]/1997
AlPO ₄	1548-1674	<u>PO</u> , <u>PO₂</u>	[231]/1997 [230]/1998
GaPO ₄	1447-1700	<u>GaPO₃</u> , <u>GaPO₂</u> , PO ₂ , PO, Ga ₂ O, O ₂	[232]/1999 [230]/1998
InPO ₄ , In(PO ₃) ₃	1200-1591	<u>InPO₃</u> , InPO ₂ , P ₄ O ₁₀ , PO, PO ₂ , In, O ₂	[230]/1998 [233]/1997
<u>TIPO₃</u> , <u>TIPO₄</u>	970-1100	TI, PO ₂ , <u>TIPO₃</u> , (TIPO ₃) ₂	[234]/1998 [230]/1998
SiP ₂ O ₇	1300-1750	PO ₂ , PO, SiO, O ₂	[235]/1995
GeP ₂ O ₇	1262-1386	P ₄ O ₁₀ , GeO, PO ₂ , <u>GePO₃</u>	[235]/1995

Table 4. contd...

Sample in the Effusion Cell	Temperature Range in K	Gaseous Species	Ref./Year
$U_{0.91}Y_{0.09}O_{1.966}$, $U_{0.88}Y_{0.12}O_{1.950}$	1800-2200	U, UO, UO ₂ , O ₂	[236]/2002
$xZnO + (1-x)P_2O_5$, $0.45 < x < 0.75$	940-1380	Zn, P ₂ , PO, PO ₂ , P ₂ O ₄ , O ₂ , ZnP ₂ O ₆	[237]/1998 [238]/1996
$x_1ZnO + x_2SnO + (1-x_1-x_2)P_2O_5$	900-1850	PO, P ₂ , PO ₂ , Zn, Sn, SnO, SnPO ₂ , SnPO ₃	[239]/1998
ZnP_2O_6 , $Zn_2P_2O_7$, $Zn_3P_2O_8$	946-1405	Zn, P ₂ , PO, PO ₂ , O	[240]/2001
$TcO_x + ReO_2$	< 1173	TcReO ₅ , TcReO ₇	[241]/1993
$LaGaO_3 + La_4Ga_2O_9$	1600-1900	Ga, Ga ₂ O, O ₂ , GaO, LaO	[242]/2001 [243]/2002
$La_{1-x}Sr_xGa_{1-y}Mg_yO_{3-(x+y)/2}$	1600-1900	Ga, Ga ₂ O, O ₂ , GaO, Sr, SrO, Mg, LaO	[242]/2001 [244]/2001 [361]/2002 [245]/2003
$LaGa_{1-x}Al_xO_3(s)$, $0 \leq x \leq 1$ $La_{0.9}Sr_{0.1}Ga_{0.8-x}Al_xMg_{0.2}O_{2.85}$, $x = 0.1, 0.2, 0.3$	1623-1928	Ga, Ga ₂ O, GaO, O ₂ , LaO, Mg, Sr, SrO	[362]/2003
$LaCrO_3 + La_2O_3$	1887-2333	Cr, CrO, CrO ₂ , LaO	[246]/1996
$LaCr_{1-x}Mg_xO_{3-\Delta}$, $x = 0.03, 0.09, 0.14$	1950-2050	Cr, CrO, CrO ₂ , Mg, LaO	[167]/2001
$La_{1-x}Ca_xCrO_{3-\Delta}$, $0.047 < x < 0.21$	1950-2050	Cr, CrO, CrO ₂ , Sr, SrO, LaO	[247]/2001
$La_{1-x}Sr_xCrO_{3-\Delta}$, $0.10 < x < 0.30$	1950-2050	Cr, CrO, CrO ₂ , CaO, LaO	[248]/2001
$La_{0.80}Sr_{0.20-x}Ca_xCrO_{3-\Delta}$, $x = 0.05, 0.10, 0.15$	1900-2100	Cr, CrO, CrO ₂ , CaO, Sr, SrO, LaO	[249]/2001
$xSnO_2 + (1-x)WO_3$, $0.50 < x < 1$	1250-1400	SnO, W ₃ O ₉ , SnWO ₄ , SnW ₂ O ₇ , Sn ₂ WO ₅ , O ₂	[250]/2002
$xBi_2O_3 + (1-x)SnO_2$, $0.10 < x < 0.975$	1000-1270	Bi, BiO, Bi ₂ O ₃ , Bi ₄ O ₆ , SnO, O ₂	[251]/2002
$Bi_2Sr_2Ca_2Cu_3O_y$, Pb-doped	650-1050	O ₂ , Bi	[252]/1995
$xTiO_2 + (1-x)Nb_2O_5$, $0.10 < x < 0.90$	2200	TiO, TiO ₂ , NbO, NbO ₂	[253]/2001
$x_1TiO_2 + x_2ZrO_2 + (1-x_1-x_2)PbO$	873-1223	PbO, Pb ₂ O ₂	[254]/1999
$xMnO + (1-x)TeO_2$, $x \geq 0.5$	950	O ₂ , Te ₂ , TeO, TeO ₂	[255]/2002
$xZrO_2 + (1-x)UO_2$, $0.55 < x < 0.98$	2200-2650	ZrO ₂ , ZrO, UO ₃ , UO ₂ , UO, O	[256]/1997
$xZrO_2 + (1-x)UO_2$, $0.05 < x < 0.70$	2000-2400	UO ₂	[257]/2001
$Mn_{0.676}Zn_{0.261}Fe_{2.063}O_4$	1250-1603	Zn, FeO, MnO	[258]/1996 [259]/1993
$xLa_2O_3 + (1-x)UO_2$, $x = 0.005, 0.01, 0.025$	2273, 2393	LaO, U, UO, UO ₂ , UO ₃ , O	[260]/2001
$Bi_2O_3 + Sb_2O_3$ in atomic ratio $Sb/(Bi+Sb) = 0.1, 0.2, 0.25, 0.3, 0.4, 0.5, 0.6, 0.666, 0.8, 0.9$	1160	Bi, BiO, Sb ₄ O ₆ , Bi ₃ SbO ₆ , Bi ₂ Sb ₂ O ₆ , BiSb ₃ O ₆ , Bi ₂ , Bi ₂ O ₃	[261]/2007
Bi-Sb-O system The enthalpies of reactions were calculated according to the second and third laws.	1100-1200	Bi, BiO, Sb ₄ O ₆ , Bi ₃ SbO ₆ , Bi ₂ Sb ₂ O ₆ , BiSb ₃ O ₆ , Bi ₂ , Bi ₂ O ₃	[262]/2009
NiO + SiO ₂ (Mo, W cell) NiO + MgHPO ₄ NiO + BaO + SiO ₂	1600-2000	NiO, XO ₂ , XO ₃ , SiO, NiXO ₄ , Ni (X=Mo,W) NiPO ₂ , NiPO ₃ , NiO, PO, PO ₂ , Ni Ni, Ba, BaO, BaNiO ₂	[263]/2006
$Na_2CO_3 + Fe_2O_3$ (2:3)	918-1023	CO ₂	[264]/2006
$K_2O + Ge_2O$ (54.80, 59.59, 62.41, 66.40, 71.83, 82.93, 91.00, 100 mol% Ge)	950-1550	K, O ₂ , GeO	[265]/2005
Mn, MnO, TeO ₂ , MnCO ₃	850-950	Te ₂ , TeO, TeO ₂ , O ₂	[266]/2006
YbSi(s), YbSi _{2-x} (s) + H ₂ (g) or H ₂ (g)/O ₂ (g)	1273-1622	H ₂ , H ₂ O, Yb, YbO, YbH, YbOH, SiO, SiO ₂	[267]/2005
PbO + O ₂ (g) GeO ₂ + O ₂ (g)	1099-1201 1220	Pb, PbO, Pb ₂ O ₂ , GeO, O ₂	[268]/2007

Table 4. contd...

Sample in the Effusion Cell	Temperature Range in K	Gaseous Species	Ref./Year
H ₃ BO ₃	295-342	<u>H₃BO₃</u>	[269]/2008
<u>TeO₂</u>	805-905	Te ₂ , <u>TeO₂</u> , <u>(TeO₂)₂</u> , <u>TeO</u> , <u>(TeO)₂</u> , O ₂	[270]/2005
SiO ₂ + H ₂ O(g)	1173, 1373, 1673	Si(OH) ₄ , SiO(OH) ₂ ,	[271]/2005
UO ₂ +O ₂ (g)	1500 and 1600	UO ₃ , O ₂	[272]/2005
U ₃ O ₈	1400-1900	O, O ₂ , U, UO, UO ₂ , UO ₃	
CeO ₂ ,	1495-1973	CeO ₂ , CeO, Sr, SrO ₂ ,	[273]/2008
SrCeO ₃ +CeO ₂ ,	1601-1988		
<u>SrCeO₃</u> ,	1643-1890		
SrCe _{0.95} Yb _{0.05} O _{2.98}	1737-1884		
U ₃ O ₈	<2800	UO, UO ₃ (I, Xe, Te, St, BaZrO ₂ , Mo, Ru, Pd, Rh)	[274]/2006
UO ₂	750-850		
Al ₂ O ₃ +C	1400-2100	Al, Al ₂ O, CO	[275]/2007
NiAl + Al ₂ O ₃	1400-1750	Al, Ni	[276]/2007
UO ₂	<2800	UO, UO ₂ , UO ₃ , Cs, BaO, SrO, Mo, Te, I, Tc, Sr, Ba,	[277]/2008
U ₃ O ₈	<1900	Rb, CsI, MoO ₃ , MoO, TeO, Te, TeO ₃	
K _{0.3} Bi _{0.5} TiO ₃	<940	K, Bi, BiO	[278]/2009
Y-Y ₂ O ₃	1423-1573	Y	[279]/2009
CaO+MgO+Al ₂ O ₃ +Cr ₂ O ₃ +FeO+SiO ₂ , FeO(0.031-0.078), MgO(0.055-0.166), CaO(0.519-0.668), Cr ₂ O ₃ (0.062-0.168), Al ₂ O ₃ (0.007-0.013), SiO ₂ (0.060-0.159)	1850-2750	CrO, CrO ₂ , Cr, Fe, SiO, Ca, Al, Mg	[280]/2009
UO ₂	1800-2700	X, XO, XO ₂ , where X= ²³⁴ U, ²³⁵ U, ²³⁶ U, ²³⁷ Np, ²³⁸ U, ²³⁸ Pu, ²³⁹ Pu, ²⁴⁰ Pu, ²⁴¹ Pu, ²⁴¹ Am, ²⁴² Pu, ²⁴³ Am, ²⁴⁴ Cm, UO ₃ , Y, YO, where Y= ¹³⁹ La, ^{140,141} Ce, ¹⁴¹ Pr, ^{143,144,145,146} Nd, ^{147,148,150,152} Sm, ¹⁵³ Eu, ¹⁵⁶ Gd, CeO ₂	[281]/2008
(CaO) ₄ P ₂ O ₅ , (CaO) ₃ P ₂ O ₅	1523-1623	P ₂ , PO	[282]/2009
Commercial PWR fuel sample (UO ₂)	< 2750	⁸⁵ Kr, ⁸⁵ Rb, UO ₂ , ¹³⁵ Cs, ²³⁸ UO ₂ , He, Xe, ⁹⁰ Sr, ¹²⁹ I, ¹³⁷ Te, ¹³⁷ Cs, ¹³⁸ Ba, ⁸⁷ Rb, ¹³⁸ BaO	[283]/2008
{LiAl ₃ O ₈ +Li ₂ CO ₃ +5Al ₂ O ₃ }, {LiAl ₃ O ₈ +2Li ₂ CO ₃ +5LiAlO ₂ }, {LiAlO ₂ +2Li ₂ CO ₃ +Li ₃ AlO ₄ },	614-750 604-723 614-750	CO ₂	[284]/2008
C	3640-3660	C, C ₂ , C ₃ , C ₅	[285]/2005
UO ₂	2500-3040	UO, UO ₂ , UO ₃	
Al (in the Knudsen cell made of α-Al ₂ O ₃ or ZrO ₂ +4%Y ₂ O ₃)	1197-1509	<u>Al</u> , <u>Al₂O</u> (ΔH sub and activities)	[286]/2006

Table 5. Summary of Mass Spectrometric Studies of Glass (Silicates and Borates) Systems. (The Gaseous Species are Underlined if their Thermodynamic Functions of Formation or of Dissociation are Given)

Sample in the Effusion Cell	Temperature Range in K	Gaseous Species	Ref./Year
SiO ₂ (Mo)	1610-1980	SiO, SiO ₂ , Si ₂ O ₂ , O	[287]/2000
SiO; Si + SiO ₂	1093-1486	SiO	[288]/1993
SiO ₂ (l) + H ₂ O	1963-2065	SiO(OH), SiO(OH) ₂ , SiO ₂	[289]/1994
x ₁ Na ₂ O + x ₂ P ₂ O ₅ + (1-x ₁ -x ₂)SiO ₂	1222-1658	NaPO ₃ , Na, O ₂	[290]/1993 [291]/1994
xNa ₂ O + (1-x)SiO ₂ , 0.195 < x < 0.651	942-1719	Na, Na ₂ O, NaO, O ₂ ,	[292]/1999 [293]/2000

Table 5. contd...

Sample in the Effusion Cell	Temperature Range in K	Gaseous Species	Ref./Year
$x_1\text{Na}_2\text{O} + x_2\text{Cs}_2\text{O} + (1-x_1-x_2)\text{SiO}_2$	950-1600	Na, Cs, O ₂	[294]/1996
$\text{K}_2\text{O} + \text{SiO}_2 (+\text{Al}_2\text{O}_3), s$	869-1050	$\text{K}_2\text{O}, (\text{Al}_2\text{O})$	[295]/2000
$x\text{K}_2\text{O} + (1-x)\text{SiO}_2, 0.108 < x < 0.500$	1057-1730	K, K ₂ O, KO, KO ₂ , O ₂	[296]/2000
$(\text{Na}_x\text{K}_{2-x})\text{Mg}_3[\text{Si}_{12}\text{O}_{30}], 0.026 < x < 0.897$	1273	Na, K	[297]/1993
$x\text{RbBO}_2 + (1-x)\text{CsBO}_2, 0 < x < 1$	764-978	$\text{RbBO}_2, \text{Rb}_2(\text{BO}_2)_2, \text{CsBO}_2, \text{Cs}_2(\text{BO}_2)_2, \text{RbCs}(\text{BO}_2)_2$	[298]/1992
$x\text{RbBO}_2 + (1-x)\text{SiO}_2, 0.077 < x < 0.87$	796-1175	$\text{RbBO}_2, (\text{RbBO}_2)_2$	[299]/1996
$x\text{CsBO}_2 + (1-x)\text{SiO}_2, 0.077 < x < 0.87$	731-1056	$\text{CsBO}_2, (\text{CsBO}_2)_2$	[300]/1999
$x\text{SiO}_2 + (1-x)\text{MgAl}_2\text{O}_4 (\text{Mo}), 0 < x < 0.7469$	1623-2523	Mg, MgO, Al, AlO, Al ₂ O, SiO, SiO ₂ , AlSiO, MoO ₂ , MoO ₃ , O	[301]/2001
$x_1\text{FeO} + x_2\text{MgO} + (1-x_1-x_2)\text{SiO}_2$	1850-1980	Fe, Mg, SiO, O ₂	[302]/1992
CaO-MgO-2SiO ₂ (+Mo)	1550-1870	Ca, Mg, SiO, SiO ₂ , MoO ₂ , MoO ₃	[303]/1996
0.51CaO + 0.49B ₂ O ₃	1415-1450	$\text{Ca}(\text{BO}_2)_2, \text{B}_2\text{O}_3$	[237]/1998
$x\text{CaO} + (1-x)\text{SiO}_2, 0.50 < x < 0.90$	1993	SiO, Ca, CaO, CaSiO ₃ , SiO ₂ , O	[304]/1991
$x_1\text{CaO} + x_2\text{Al}_2\text{O}_3 + (1-x_1-x_2)\text{SiO}_2 (+ \text{Mo})$	1833-2033	Al, AlO, Al ₂ O, SiO, SiO ₂ , MoO ₂ , MoO ₃	[305]/1996
$x_1\text{CaO} + x_2\text{Al}_2\text{O}_3 + (1-x_1-x_2)\text{SiO}_2 (+ \text{Nb}, \text{Ta}, \text{Mo})$	1634-1841	SiO, NbO ₂ , MoO ₂ , TaO ₂	[306]/1998
$x_1\text{CaO} + x_2\text{TiO}_2 + (1-x_1-x_2)\text{SiO}_2$	1850-2200	SiO, O	[307]/2000
$x\text{BaO} + (1-x)\text{B}_2\text{O}_3, x = 0.333, 0.50$	1277-1535	Ba, <u>BaBO</u> , BO, B ₂ O ₂ , B ₂ O ₃	[308]/1990
SrO + B ₂ O ₃ + SiO ₂	1337-1602	Sr, <u>SrSiO</u> , <u>SrBO</u> , BO, B ₂ O ₂ , B ₂ O ₃ , SiO	[309]/1990
BaSiO ₃	1783-2143	Ba, BaO, SiO, SiO ₂ , <u>BaSiO</u> ₃	[310]/2000
$x\text{B}_2\text{O}_3 + (1-x)\text{SiO}_2, 0.224 < x < 0.896$	1450-1500	B ₂ O ₃	[311]/1993
B ₂ O ₃ + SiO ₂ + other oxides	1273-1973	B ₂ O ₃ , SiO, Si, O ₂ , O, Al, Al ₂ O, AlO, XBO ₂ , X = Li, Na, K	[312]/1990
B ₂ O ₃ + SiO ₂ + other oxides	900-1650	B ₂ O ₃ , SiO, O ₂ , Pb, XBO ₂ , X ₂ (BO ₂) ₂ , XY(BO ₂) ₂ , X, Y = Li, Na, K, Cs	[313]/1998
$x\text{Al}_2\text{O}_3 + (1-x)\text{B}_2\text{O}_3, 0.92 < x < 0.50$	1248-1850	B ₂ O ₃	[314]/1995
<u>3Al₂O₃:2SiO₂</u> (+ Mo)	1833-2033	AlO, Al ₂ O, SiO, SiO ₂ , <u>AlSiO</u> , MoO ₂ , MoO ₃ , O, O ₂	[315]/1994
$x\text{Al}_2\text{O}_3 + (1-x)\text{SiO}_2, 0.0075 < x < 0.850$	1720-2520	SiO, Al, AlO, O	[316]/2000
$x\text{Al}_2\text{O}_3 + (1-x)\text{SiO}_2, 0.10 < x < 0.60$ (in W)	1863-2169	SiO, WO ₃ , WO ₂	[317]/2001
$x\text{Al}_2\text{O}_3 + (1-x)\text{SiO}_2 (\text{Mo}), 0.0119 < x < 0.8414$	1676-2327	SiO, Al, O	[318]/2002 [319]/2003 [320]/2003
Al ₂ O ₃ + SiC + C + Si	1350-1750	Al, Al ₂ O, CO, SiO	[321]/2003 [322]/2003
$x\text{MnO} + (1-x)\text{SiO}_2, 0.40 < x < 0.80$	1369-1817	Mn, SiO	[323]/1995
$x\text{BaO} + (1-x)\text{SiO}_2, 0.2 < x < 0.6$	1838-1990	SiO, BaO, Ba, MoO ₂ , MoO ₃ , BaMoO ₃ , BaMoO ₄	[324]/2006
<u>Al₁₈B₄O₃₃</u> + Al ₂ O ₃	1573-1673	B ₂ O ₃	[325]/2009
Al ₁₈ B ₄ O ₃₃ + B ₂ O ₃ rich liquid	1373-1423		
B ₂ O ₃ rich liquid	1373-1423		
SiC+SiO ₂	< 1650 K	SiO, CO	[326]/2008

Table 6. Summary of Mass Spectrometric Studies on the Metal Halide-Metal Oxide Systems

Sample in the Effusion Cell	Temperature Range in K	Gaseous Species	Ref./Year
CaO + CaF ₂	800-2240	CaF ₂ , Ca, O ₂	[327]/1998
$x_1\text{CaO} + x_2\text{SiO}_2 + (1-x_1-x_2)\text{CaF}_2 (+ \text{Nb}, \text{Ta}, \text{Mo})$	1370-1849	SiF ₄ , SiF ₂ , CaF ₂ , CaF, Ca, SiO, MO ₂ , (M = Nb, Ta, Mo)	[328]/1992 [329]/1992

Table 6. contd...

Sample in the Effusion Cell	Temperature Range in K	Gaseous Species	Ref./Year
$x_1\text{CaO}+x_2\text{SiO}_2+x_3\text{Al}_2\text{O}_3+(1-x_1-x_2-x_3)\text{CaF}_2$ (+Nb, Ta, Mo)	1370-1836	$\text{SiF}_4, \text{SiF}_2, \text{CaF}_2, \text{CaF}, \text{Al}_2\text{F}_6, \text{AlF}_3, \text{AlF}_2, \text{AlF}, \text{AlOF}, \text{Ca}, \text{SiO}, \text{MO}$ (M = Nb, Ta, Mo)	[330]/1997
$\text{CaO} + \text{CaCl}_2$ (+ Al, Al_2O_3)	800-2240	$\text{CaCl}_2, \text{AlOCl}, \text{Ca}, \text{HCl}$	[327]/1998
$\text{CaO} + \text{CaBr}_2$	1246	$\text{CaBr}_2, \text{Ca}_2\text{Br}_4, \text{Ca}_3\text{OBr}_4, \text{Ca}_4\text{OBr}_6$	[331]/1992
$\text{CaO} + \text{CaI}_2$	1048	$\text{CaI}_2, \text{Ca}_2\text{I}_4, \text{Ca}_3\text{OI}_4, \text{Ca}_4\text{OI}_6$	[331]/1992
$x\text{Dy}_2\text{O}_3 + (1-x)\text{DyF}_3, 0.05 < x < 0.90$	1373	$\text{DyF}_3, \text{DyOF}, \text{Dy}_2\text{OF}_4,$	[332]/1998
$\text{UO}_{2+x} + \text{CsI}, 0.00 < x < 0.29$	600-2300	$\text{I}_2, \text{CsI}, \text{Cs}_2\text{I}_2, \text{Cs}, \text{UO}_2,$	[333]/1998
$(\text{AgI})_{0.5}-(\text{AgPO}_3)_{0.5}$	500-650	$\text{I}, \text{PO}, \text{P}_4\text{O}_{10}, \text{AgI}$	[334]/2009
$\text{TmX}_3+\text{NaX}+\text{Al}_2\text{O}_3$ (X=I,Br)	744-1134 (X=Br) 842-1103 (X=I)	$\text{NaX}, \text{Na}_2\text{X}_2, \text{TmX}_3, \text{Tm}_2\text{X}_6, \text{Na}_2\text{TmX}_5, \text{AlX}_3, \text{NaAlX}_4$	[335]/2005
Mould powders	373-1823	$\text{NaF}, \text{Na}_2\text{F}_2, \text{KF}, \text{SiF}_4, \text{CaF}_2, \text{AlF}_3, \text{MgF}_2, \text{AlOF}, \text{Na}, \text{SiO}, \text{Mg}, \text{MgF}, \text{SiF}_2$	[336]/2005

Table 7. Summary of Mass Spectrometric Studies of Fullerenes and their Derivatives

Sample in the Effusion Cell	Temperature Range in K	Gaseous Species	Ionisation Properties	Thermodynamic Functions	Ref./Year
C_{60}	757	C_{60}	$\text{C}_{60}^+/ \text{C}_{60}$ $\text{C}_{60}^{2+}/ \text{C}_{60}$	-	[337]/1996
C_{60}	773-873	C_{60}	$\text{C}_{54}^+/ \text{C}_{60}$ $\text{C}_{56}^+/ \text{C}_{60}$	-	[338]/1996
C_{60}	600-800	C_{60}	$\text{C}_{60}^+/ \text{C}_{60}$	$p(i), \Delta_{\text{sub}}H$	[339]/1992 [340]/1994
C_{60}		C_{60}	$\text{C}_{60}^{k+}, \text{C}_{58}^{l+}, \text{C}_{56}^{m+}, \text{C}_{54}^{n+}, \text{C}_{50}^{m+}, \text{C}_{48}^{m+}, \text{C}_{46}^+, \text{C}_{44}^+,$ $k=1-4, l=1-3, m=1,2$		[341]/1996
C_{70}		C_{70}	$\text{C}_{70}^{k+}, \text{C}_{68}^{k+}, \text{C}_{66}^{k+}, \text{C}_{64}^{k+}, \text{C}_{60}^{k+}, \text{C}_{62}^{m+}, \text{C}_{58}^{m+},$ $\text{C}_{56}^{n+}, \text{C}_{54}^{n+}, \text{C}_{52}^{2+}, \text{C}_{50}^{2+}, k=1-4, m=1-3, n=2,3$		[341]/1996
C_{70}	843	C_{70}	$\text{C}_{70}^+/ \text{C}_{70}$ $\text{C}_{70}^{2+}/ \text{C}_{70}$	-	[337]/1996
C_{70}	650-850	C_{70}	$\text{C}_{70}^+/ \text{C}_{70}$	-	[342]/1992
C_{70}	650-850	C_{70}	-	$p(i), \Delta_{\text{sub}}H$	[343]/1993
$\text{C}_{60} + \text{C}_{70}$	700-840	$\text{C}_{60}, \text{C}_{70}$	-	$p(i), a(i), \Delta_{\text{mix}}G$	[344]/1994 [340]/1994
$\text{C}_{60} + \text{C}_{70}$	795-1006	$\text{C}_{70+2n}, n=0-18$		electron affinities	[345]/1996
C_{76}	637-911	C_{76}	-	$p(i), \Delta_{\text{sub}}H$	[346]/1998
C_{84}	658-980	C_{84}		$p(i), \Delta_{\text{sub}}H$	[347]/1998
C_n^- mixtures	886-983	$\text{C}_{2n}, n=35-53$	-	$\text{C}_{2n} + \text{C}_{60}^+ = \text{C}_{60} + \text{C}_{2n}^+$	[348]/2000
C_n^- mixtures	750-1050	$\text{C}_{70+2n}, n=0-18$		electron affinities	[345]/1996 [349]/1997
C_n^- mixtures	700-950	$\text{C}_{76}, \text{C}_{78}, \text{C}_{84}$		$p(i), \Delta_{\text{sub}}H$	[347]/1998
$\text{C}_{60}\text{H}_n, n=18, 36$	550-685	C_{60}H_n		$p(i), \Delta_{\text{sub}}H$	[350]/2001
$\text{C}_{60}\text{F}_2, \text{C}_{70}\text{F}_2$	650-1070	$\text{C}_{60}\text{F}_2, \text{C}_{70}\text{F}_2, \text{C}_n\text{F}_2^+, n=60, 70, 72, 74, 76, 78$		electron affinities	[351]/1994

Table 7. contd...

Sample in the Effusion Cell	Temperature Range in K	Gaseous Species	Ionisation Properties	Thermodynamic Functions	Ref./Year
C ₆₀ F ₃₆	408-539	C ₆₀ F ₃₆ , C ₆₀ F ₃₄ O		p(i), Δ _{sub} H	[352] [353]
C ₆₀ F ₄₈	395-528	C ₆₀ F ₄₈		p(i), Δ _{sub} H	[352]
KF + Gd@C _n	931-960	C ₆₀ , C _{70+2n} , (n = 0-10),		electron affinities	[354]
C ₆₀ + NiF ₂	500-800	C ₆₀ , C ₆₀ F _{2n} , n = 1-23			[355]
C ₆₀ + BaPbF ₆ , MgPbF ₆ , Na ₃ PbF ₇ , K ₃ PbF ₇ , MnF ₃ , CoF ₃	500-800	C ₆₀ , C ₆₀ F _n , n=18-46		Δ _r H, kinetics	[356]
C ₆₀ /C ₇₀ + CrCl ₃ , FeCl ₃ , CuCl ₂	520-720	C ₆₀ Cl _{2n} (n= 0-9) C ₇₀ Cl _{2m} (m = 0-7)		Δ _r H	[357]
K + C ₆₀	560-960	K, C ₆₀ , K@C ₆₀		Δ _r H, phase diagram	[358] [356]
C ₆₀ +KMnF ₄ , K ₃ MnF ₆ , KMnF ₅ , A ₃ MnF ₆ (A=Li, K, Cs)	320-480	C ₆₀ , C ₆₀ F _n (n = 8, 18,20,36,38,40)		C ₆₀ F ₈	[359]
C ₆₀ F ₁₈ , C ₆₀ F ₃₆ , C ₆₀ F ₄₈ C ₆₀ (CF ₃) _n (n = 4-16) C ₆₀ + K ₃ PbF ₆ , K ₂ Pf ₆ , COF ₃ and SFAs	550-876	C ₆₀ F _n (n=8-40)	Different negative ions	electron affinity	[360]

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